

Supplementary Materials

Synergistic promotion of ultra-small Pt nanoparticles and oxygen vacancy in MOF catalyst for ethyl levulinate to valerolactone at room temperature

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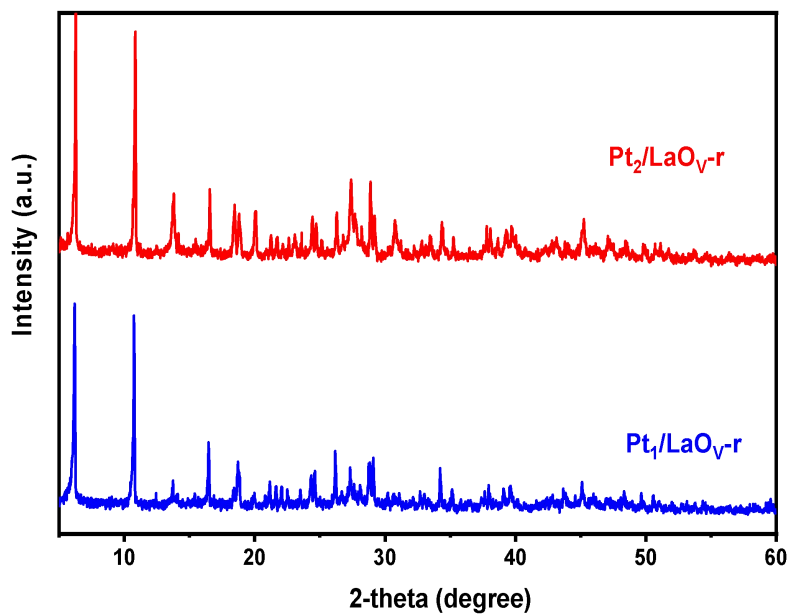
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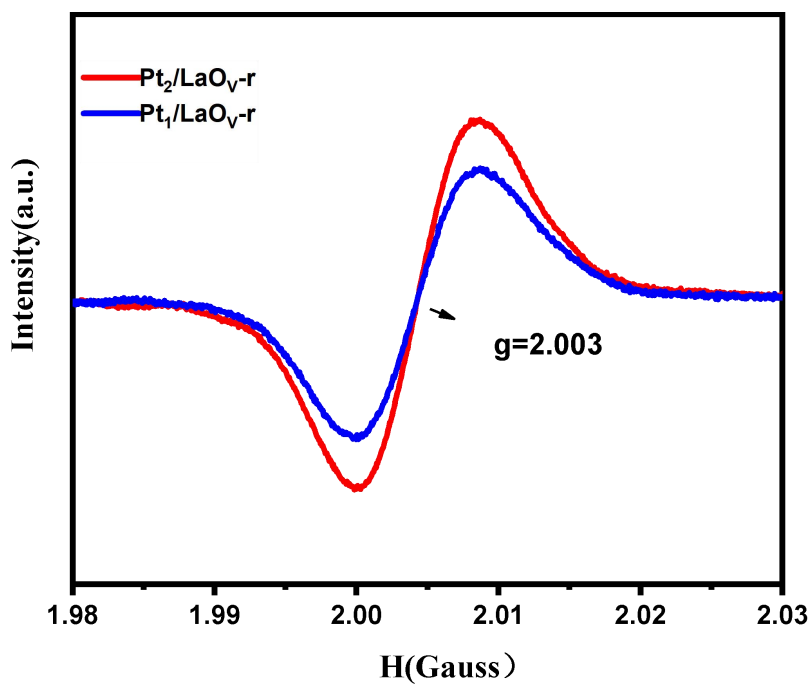
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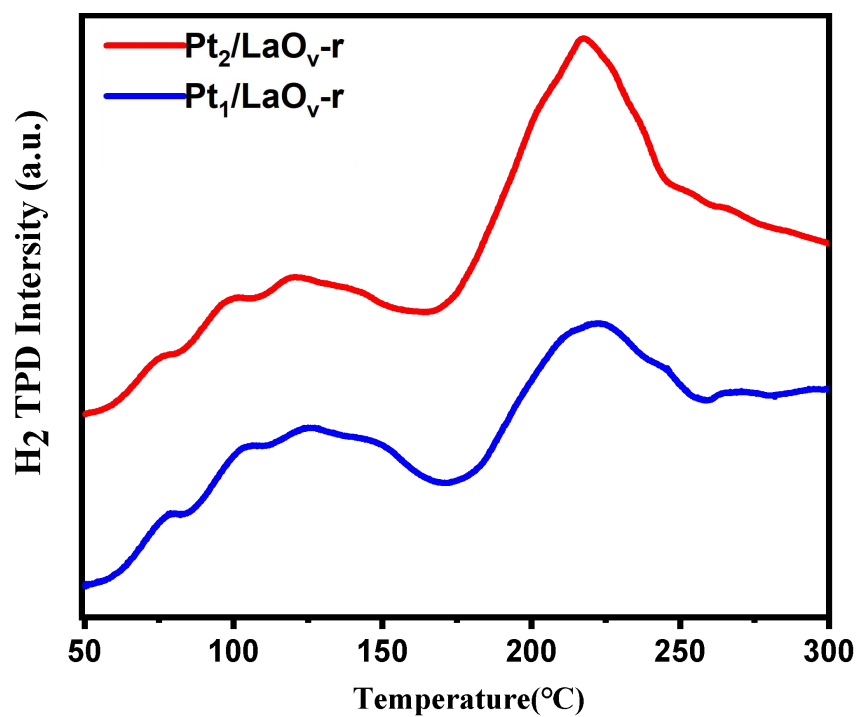
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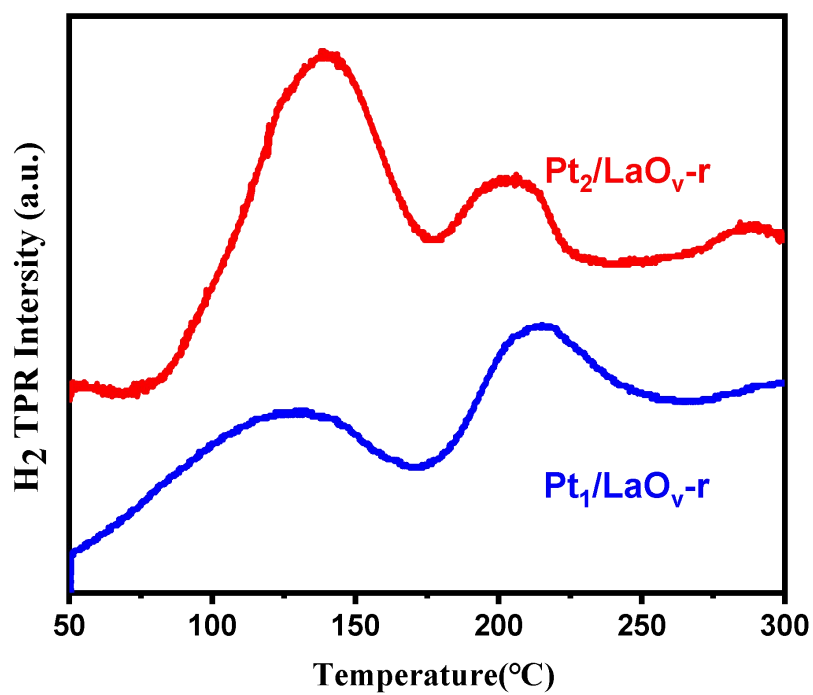
Supplementary Figure 1. XRD patterns of Pt₁/LaO_v-r and Pt₂/LaO_v-r.



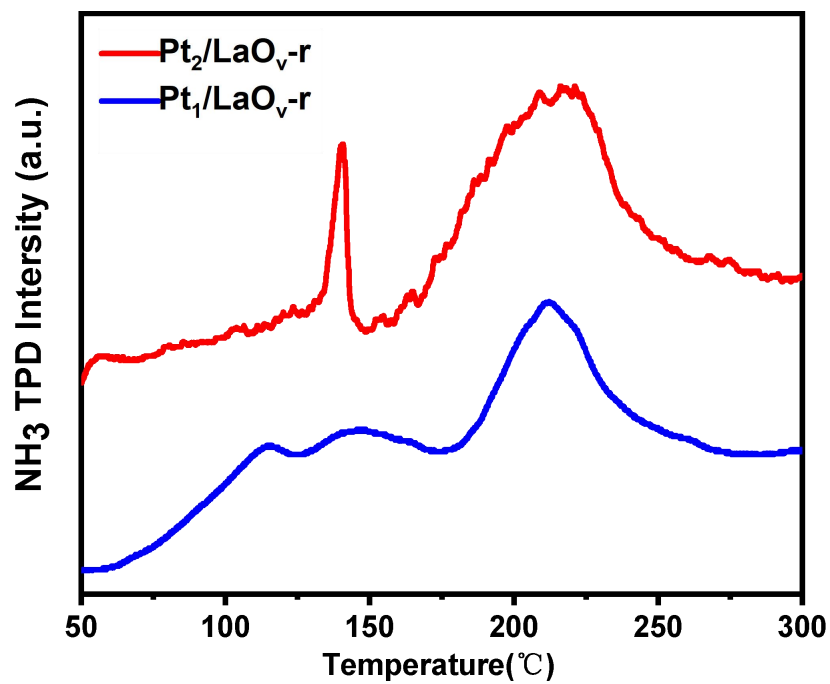
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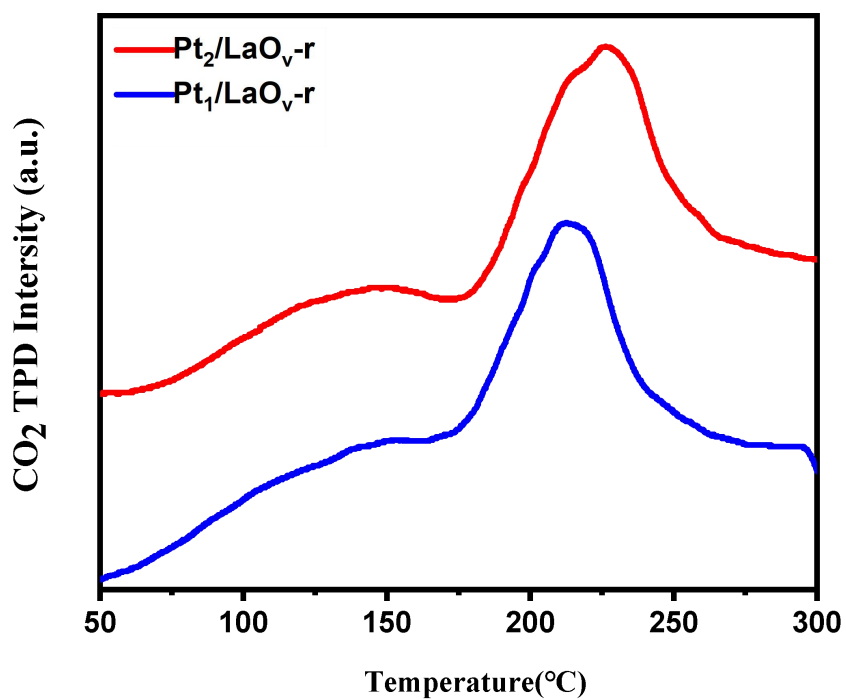
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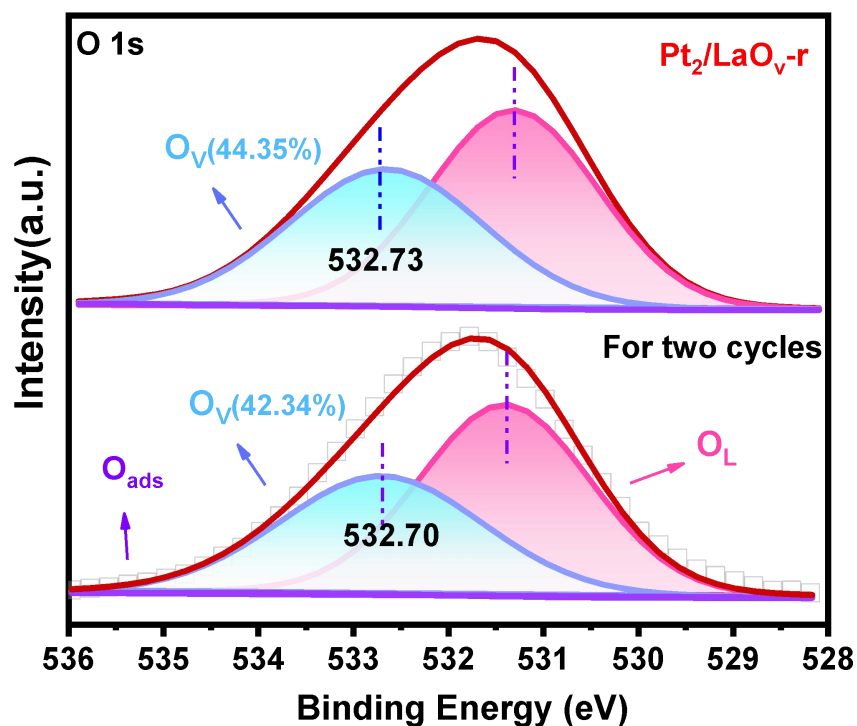
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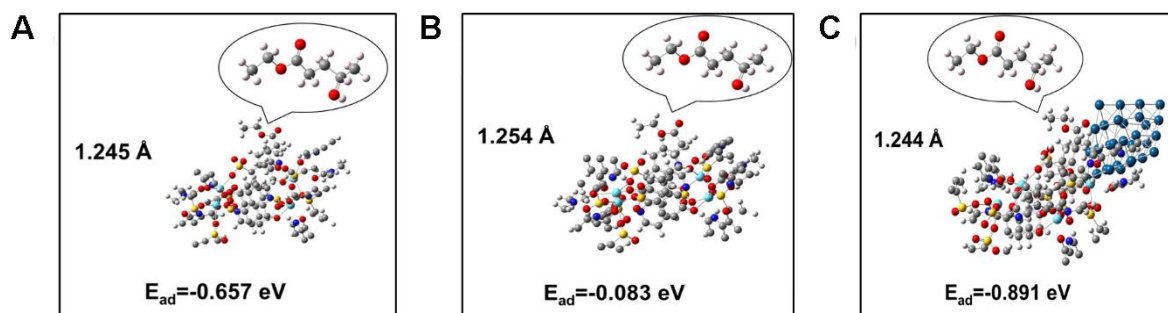
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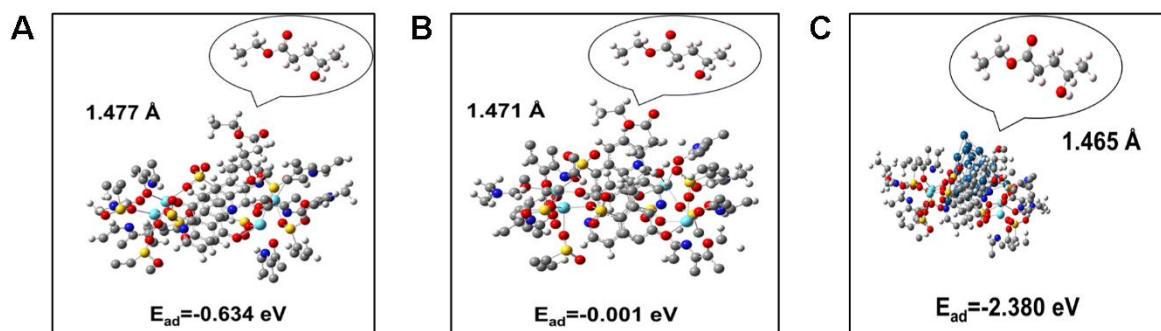
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Supplementary Figure 7. XPS spectra of O 1s in Pt_2/LaO_{V-r} after cycling.



Supplementary Figure 8. (A-C) EL adsorbed on La and O_V in LaO_{V-r} , La in Pt_2/LaO_{V-r} .



Supplementary Figure 9. (A-C) Intermediates adsorbed on La and O_V in LaO_{V-r} , Pt in $Pt_2/LaQS$.

Supplementary Table 1. the comparison results of the catalytic properties with the existing literature

Catalysts	T (°C)	P (MPa)	t (h)	Solvent	Subst rate	Product	Con. (%)	Select. (%)	Ref.
Pt-TiO ₂ /α-Al ₂ O ₃	150	3	3	H ₂ O	EL	GVL	98	98	[1]
Pt/Sn _{0.8} Mn ₁ O _y	120	2	6	Dioxane	LA	GVL	99	>99	[2]
Pt _{0.7} -Ce _{0.5} /TS-1	180	2	6	H ₂ O	LA	GVL	99	>99	[3]
1.6Pt/mesoporous ZrO ₂	240	2.5	24	H ₂ O	LA	GVL	90	90	[4]
Pt ₂ /LaO _{v-r}	80	2	4	2-PrOH	EL	GVL	>99	>99	This Work
Pt ₂ /LaO _{v-r}	25	2	36	2-PrOH	EL	GVL	>99	>99	This Work

Supplementary Table 2. Acid contents of different catalysts measured by NH₃-TPD

Entry	Catalyst	NH ₃ -TPD acid content (μmol/g)	Medium acid content (μmol/g)
1	Pt ₂ /LaO _{v-r}	3375.6	1803.6
2	Pt ₁ /LaO _{v-r}	2944.5	1561.5
4	Pt ₂ /LaQS	3248.0	1417.4
5	LaO _{v-r}	2047.8	376.6
6	LaQS	1992.4	680.9

Supplementary Table 3. Base contents of different catalysts measured by CO₂-TPD

Entry	Catalyst	CO ₂ -TPD base content (μmol/g)	Medium base content (μmol/g)
1	Pt ₂ /LaO _{v-r}	2350.5	1554.6
2	Pt ₁ /LaO _{v-r}	2022.3	1133.9
4	Pt ₂ /LaQS	1707.3	812.1
5	LaO _{v-r}	1189.4	459.4
6	LaQS	387.6	322.3

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