Supplementary Materials

Data-driven strategy for bandgap database construction of perovskites and the potential segregation study

Bobin Wu^{1,2,3}, Xinyu Zhang^{1,2}, Zixuan Wang^{1,2}, Zijian Chen^{1,2}, Shaohui Liu^{1,2,3}, Jie Liu⁴, Zhenming Xu^{5,*}, Qingde Sun^{6,7,*}, Haitao Zhao^{1,2,*}

¹Center for Intelligent and Biomimetic Systems, Shenzhen Institute of Advanced Technology, Chinese Academy of Sciences, Shenzhen 518055, Guangdong, China.
²Digital Intelligent Manufacturing Research Center, Wenzhou Institute of Technology of the Chinese Academy of Sciences, Wenzhou, 325000, Zhejiang, China.
³Nano Science and Technology Institute, University of Science and Technology of China, Suzhou 215000, Jiangsu, China.

⁴Department of Chemistry, The University of Hong Kong, Hong Kong 999077, China. ⁵Jiangsu Key Laboratory of Electrochemical Energy Storage Technologies, College of Materials Science and Technology, Nanjing University of Aeronautics and Astronautics, Nanjing 210016, Jiangsu, China.

⁶School of Physics and Electronic Science, Changsha University of Science and Technology, Changsha 410114, Hunan, China.

⁷School of Materials Science & Engineering, Nanyang Technological University, Singapore 639798, Singapore.

*Correspondence to: Dr. Haitao Zhao, Center for Intelligent and Biomimetic Systems, Shenzhen Institute of Advanced Technology, Chinese Academy of Sciences, 1068 Xueyuan Avenue, Shenzhen University Town, Shenzhen 518055, Guangdong, China. E-mail: haitaozhao@outlook.com; Dr. Zhenming Xu, Jiangsu Key Laboratory of Electrochemical Energy Storage Technologies, College of Materials Science and Technology, Nanjing University of Aeronautics and Astronautics, No. 29 Yudao Street, Nanjing 210016, Jiangsu, China. E-mail: xuzhenming@nuaa.edu.cn; Dr. Qingde Sun, School of Materials Science & Engineering, Nanyang Technological University, 50 Nanyang Ave, Singapore 639798, Singapore. E-mail: qingde.sun@ntu.edu.sg

Compounds	Bandgap	Ref
Cubic FAPb(I _{1-x} Br _x) ₃	1.48(1 - x) + 2.23x - 0.15x(1 - x)	[1]
Cubic MAPb(I _{1-x} Br _x) ₃	1.57(1 - x) + 2.29x - 0.33x(1 - x)	[2]
Cubic CsPb(I _{1-x} Br _x) ₃	1.77(1 - x) + 2.38x - 0.35x(1 - x)	[3]
Tetragonal MAPb(I _{1-x} Br _x) ₃	1.57(1 - x) + 2.25x - 0.3x(1 - x)	[4]
Orthorhombic CsPb(I _{1-x} Br _x) ₃	1.73(1 - x) + 2.34x - 0.22x(1 - x)	[5]

Supplementary Table 1. Formulas for the bandgap in eV as a function of relative Br concentration x for the five compounds



Supplementary Figure 1. DFT calculated electronic band structures for six representative mixed halide perovskites in cubic structure. (A) FA_{0.75}Cs_{0.25}PbI₃; (B) FA_{0.5}MA_{0.5}PbBr₃; (C) FA_{0.5}MA_{0.5}Pb(I_{0.33}Br_{0.67})₃; (D) FA_{0.5}MA_{0.5}Pb(I_{0.5}Br_{0.5})₃; (E) FA_{0.5}MA_{0.5}Pb(I_{0.67}Br_{0.33})₃; (F) FA_{0.5}MA_{0.5}Pb(I_{0.83}Br_{0.17})₃.



Supplementary Figure 2. Performances of fine-tuned MEGNet in different datasets. (A) DFT bandgap dataset; (B) The experimental bandgap dataset; (C) The gap between experimental and DFT bandgap dataset.



Supplementary Figure 3. Total energy prediction results of different graph neural networks. (A) M3GNet; (B) CHGNet.



Supplementary Figure 4. Prediction results of different methods in DFT bandgap dataset. (A) Pretrained MatGL; (B) Pretrained MEGNet; (C) DFT method.



Supplementary Figure 5. The probability density distribution of perovskite bandgap database.



Supplementary Figure 6. (A) Decomposition energy; (B) Mixing energy; (C) Formation energy; (D) Energy above hull of FAPb(I_{1-x}Br_x)₃.



Supplementary Figure 7. ΔE_g for the different compounds (APb(I_{1-x}Br_x)₃). A = (A) FA_{0.75}Cs_{0.25}, FA_{0.5}Cs_{0.5} and FA_{0.25}Cs_{0.75}; (B) MA_{0.75}Cs_{0.25}, MA_{0.5}Cs_{0.5} and MA_{0.25}Cs_{0.75}; (C) FA_{0.75}MA_{0.25}, FA_{0.5}MA_{0.5} and FA_{0.25}MA_{0.75}.

REFERENCES

1. Eperon GE, Stranks SD, Menelaou C, Johnston MB, Herz LM, et al. Formamidinium lead trihalide: a broadly tunable perovskite for efficient planar heterojunction solar cells. *Energy & Environmental Science* 2014;7:982. https://doi.org/10.1039/c3ee43822h

2. Noh JH, Im SH, Heo JH, Mandal TN, Seok SI. Chemical Management for Colorful, Efficient, and Stable Inorganic–Organic Hybrid Nanostructured Solar Cells. *Nano Letters* 2013;13:1764-9. <u>https://doi.org/10.1021/nl400349b</u>

3. Beal RE, Slotcavage DJ, Leijtens T, Bowring AR, Belisle RA, et al. Cesium Lead Halide Perovskites with Improved Stability for Tandem Solar Cells. *The Journal of Physical Chemistry Letters* 2016;7:746-51. <u>https://doi.org/10.1021/acs.jpclett.6b00002</u>

4. Atourki L, Vega E, Marí B, Mollar M, Ahsaine HA, et al. Role of the chemical substitution on the structural and luminescence properties of the mixed halide perovskite thin MAPbI3–xBrx ($0 \le x \le 1$) films. *Applied Surface Science* 2016;371:112-7. <u>https://doi.org/10.1016/j.apsusc.2016.02.207</u>

5. Jung J, Yun Y, Yang SW, Oh HG, Jeon A-Y, et al. Ternary diagrams of phase, stability, and optical properties of cesium lead mixed-halide perovskites. *Acta Materialia* 2023;246:118661. <u>https://doi.org/10.1016/j.actamat.2022.118661</u>