

Supplementary Materials

Modulating the graphitic domains of hard carbons via tuning resin crosslinking degree to achieve high rate and stable sodium storage

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We report a hard carbon (HC) material derived from 3-aminophenol formaldehyde resin (AFR) with better rate performance, higher initial coulombic efficiency (ICE) and capacity via adjusting the crosslinking degree of the AFR precursor. We discover that a resin precursor with a higher crosslinking degree tends to produce HC with higher carbon yield, a larger d_{002} , fewer defects, and a lower surface area.

Supplementary Table 1. Physical parameters for 0.46-AFR-HC, 0.90-AFR-HC and 1.80-AFR-HC

		0.46-AFR-HC	0.90-AFR-HC	1.80-AFR-HC
	d_{002} (nm)	0.388	0.384	0.372
TEM	L_c (nm)	1.40	1.36	1.32
	L_a (nm)	2.64	2.59	2.56
	d_{002} (nm)	0.401	0.389	0.374
XRD	L_c (nm)	1.46	1.42	1.33
	L_a (nm)	3.01	2.89	2.88

$d_{002} = n\lambda/2\sin\theta = 0.154187/2\sin\theta$, $L = K\gamma/B\cos\theta$, $N = Lc/d_{002}+1$, K is the Scherrer constant (usually set as 0.98), γ is the wavelength of the X-ray beam, B is the full width at half maximum (FWHM) value of the (002) and (100) peak, 2θ is the diffraction angle, and L is the average thickness in the per-pendicular direction of the crystal plane. L_a : the apparent layer-plane length parallel to the fiber axis; L_c : the apparent crystallite thickness; d_{002} : the average interlayer distance^[1-3].

Supplementary Table 2. Raman parameters for 0.46-AFR-HC, 0.90-AFR-HC and 1.80-AFR-HC

		A_D	A_G	A_D/A_G
	0.46-AFR-HC	85.43817	60.85909	1.4039
	0.90-AFR-HC	95.1652	58.9377	1.6146
	1.80-AFR-HC	104.3127	60.7502	1.7171
Raman		I_D	I_G	I_D/I_G
	0.46-AFR-HC	0.9491	0.9968	0.9521
	0.90-AFR-HC	0.9723	0.9815	0.9906
	1.80-AFR-HC	0.9974	0.956	1.0433

Supplementary Table 3. Pore structure and specific surface area information for 0.46-AFR-HC, 0.9-AFR-HC and 1.80-AFR-HC

N₂	BET surface area (m²/g)	CO₂	Micropore surface area (m²/g)
0.46-AFR-HC	11.1750	0.46-AFR-HC	506.4595
0.90-AFR-HC	35.7286	0.90-AFR-HC	518.7026
1.80-AFR-HC	52.8906	1.80-AFR-HC	634.4605

Supplementary Table 4. Atomic ratios of elements C, N, and O in 0.46-AFR-HC, 0.90-AFR-HC and 1.80-AFR-HC from XPS

	C (at%)	N (at%)	O (at%)
0.46-AFR-HC	93.29	1.27	5.36
0.90-AFR-HC	92.63	1.28	5.98
1.80-AFR-HC	92.07	1.44	6.4

Supplementary Table 5. The compaction density of 0.46-AFR-HC, 0.90-AFR-HC and 1.80-AFR-HC

Compaction density (g/cm³)	
0.46-AFR-HC	0.82
0.90-AFR-HC	0.79
1.80-AFR-HC	0.75

Supplementary Table 6. Summary of representative publications with high capacity of HC and other anode materials previously reported in the literature

Materials	Capacity (mAh·g⁻¹)	Rate capability (mAh·g⁻¹)	Temperature (°C)	Ref.
Seedpod-HC	328	85 @ 1 A·g ⁻¹	1,200	[4]
Pitch-C	250	124 @ 10 A·g ⁻¹	1,000	[5]
O-carbon	380	152 @ 2 A·g ⁻¹	1,600	[6]
N-HC	364	214 @ 10 A·g ⁻¹	1,200	[7]
HC	320	80 @ 5 C	1,200	[8]
Pitch and resin-HC	350	141 @ 20 A·g ⁻¹	1,000	[9]
Cu-carbon microcuboid	201	136 @ 2 A·g ⁻¹	600	[10]
0.46-AFR-HC	383	140 mAh·g ⁻¹ @ 20 A·g ⁻¹	1,300	This work

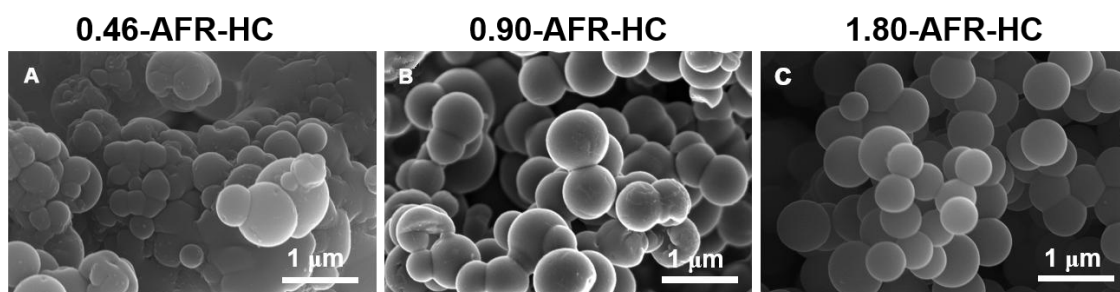
Supplementary Table 7. Summary of representative publications with high ICE of HC previously reported in the literature

Materials	ICE (%)	Electrolyte	Ref.
HCNS	82	1 M NaPF ₆ in diglyme	[7]
HC-1	79	1 M NaPF ₆ in TEGDME	[8]
Pitch and resin-HC	63	1 M NaPF ₆ in diglyme	[9]
Cu-carbon microcuboid	60	1 M NaCF ₃ SO ₃ in diglyme	[10]
HC-2	75	1 M NaPF ₆ in diglyme	[2]
0.46-AFR-HC	82	1 M NaPF ₆ in diglyme	This work

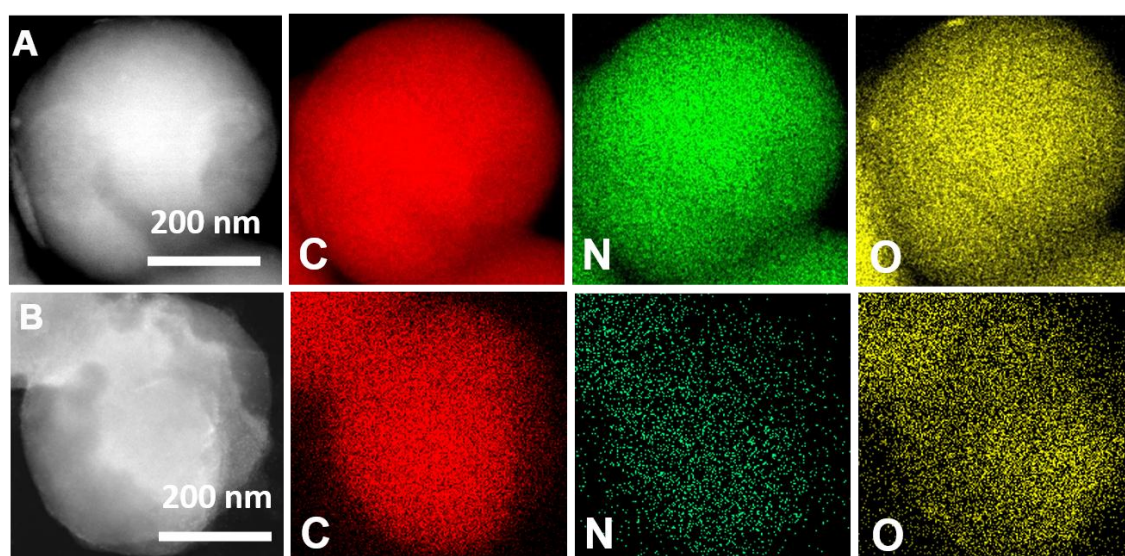
Supplementary Table 8. EIS fitting data for 0.46-AFR-HC, 0.90-AFR-HC and 1.80-AFR-HC electrodes

Samples	Cycles	R_s	R_{SEI}	R_{ct}
0.46-AFR-HC	10	10.3	2.6	3.6
0.90-AFR-HC	10	10.4	3.9	7.9
1.80-AFR-HC	10	10.4	7.3	11.8

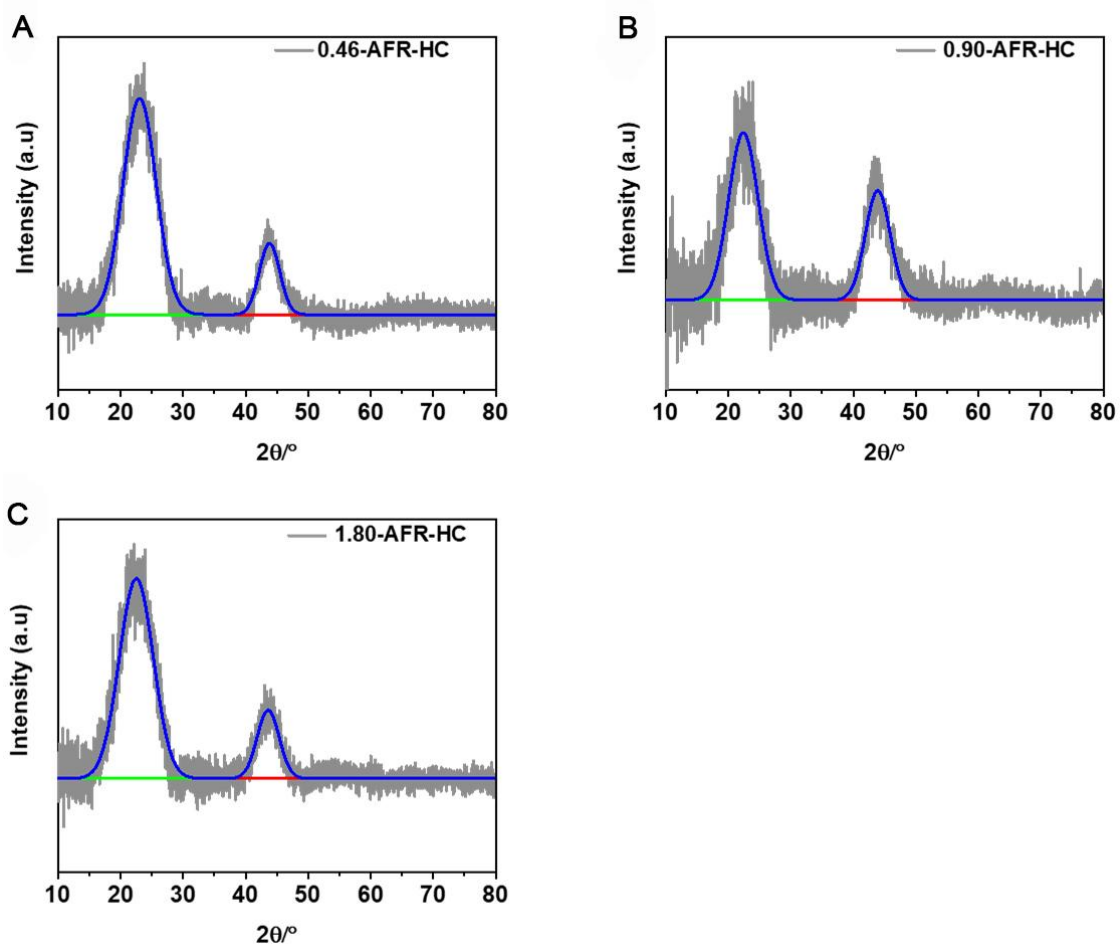
R_s: Internal resistance including the resistances of the electrolyte solution and electrodes; R_{SEI}: interfacial resistance of electrode due to formation of SEI; R_{ct}: resistance associated with charge transfer.



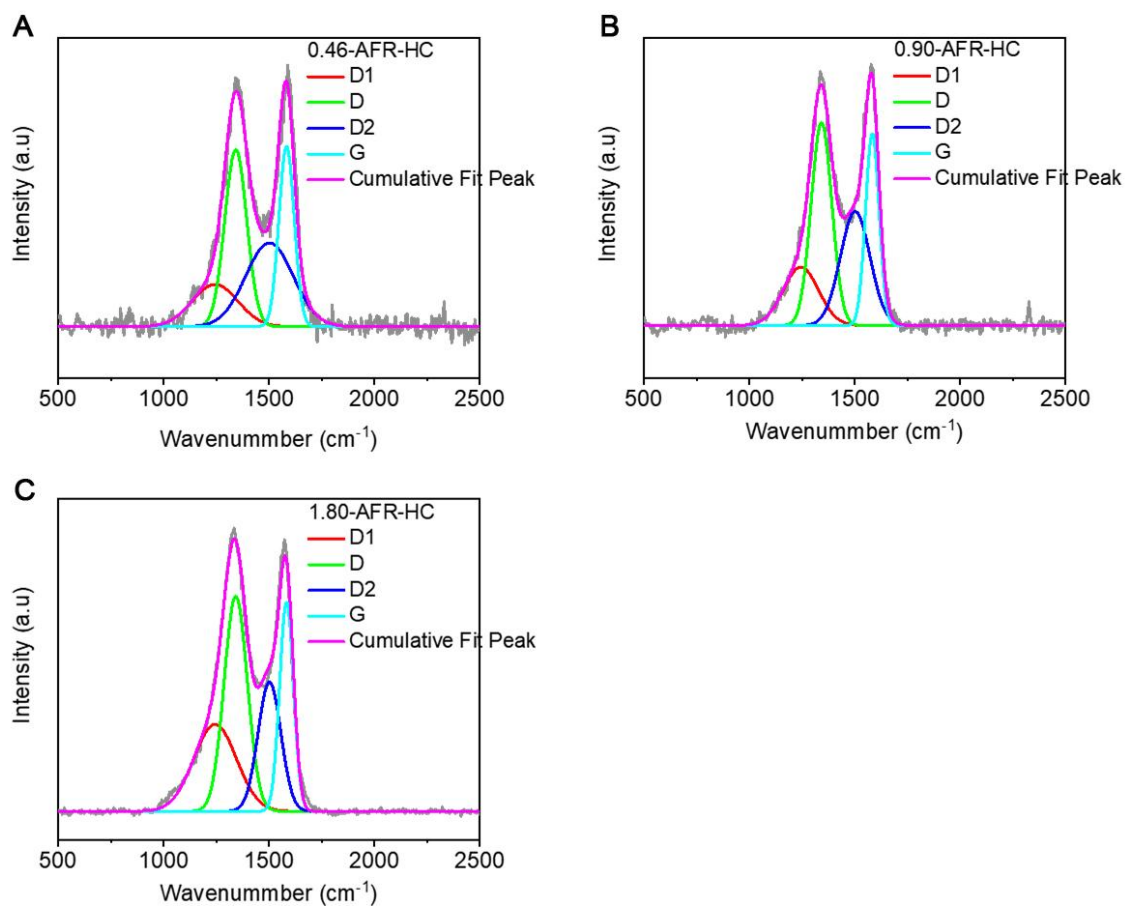
Supplementary Figure 1. SEM images for (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; and (C) 1.80-AFR-HC.



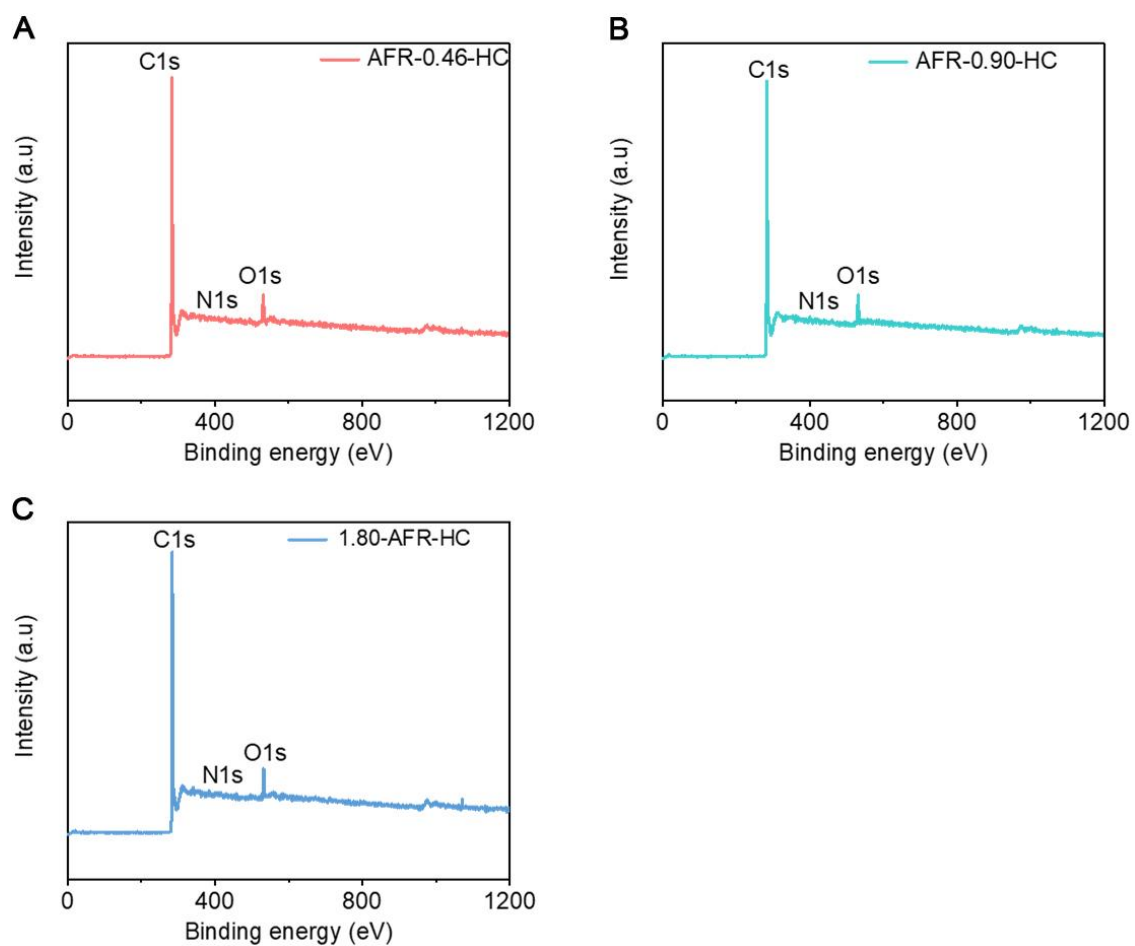
Supplementary Figure 2. Energy-dispersive X-ray spectroscopy mapping images for (A) 0.46-AFR-HC and (B) 1.80-AFR-HC.



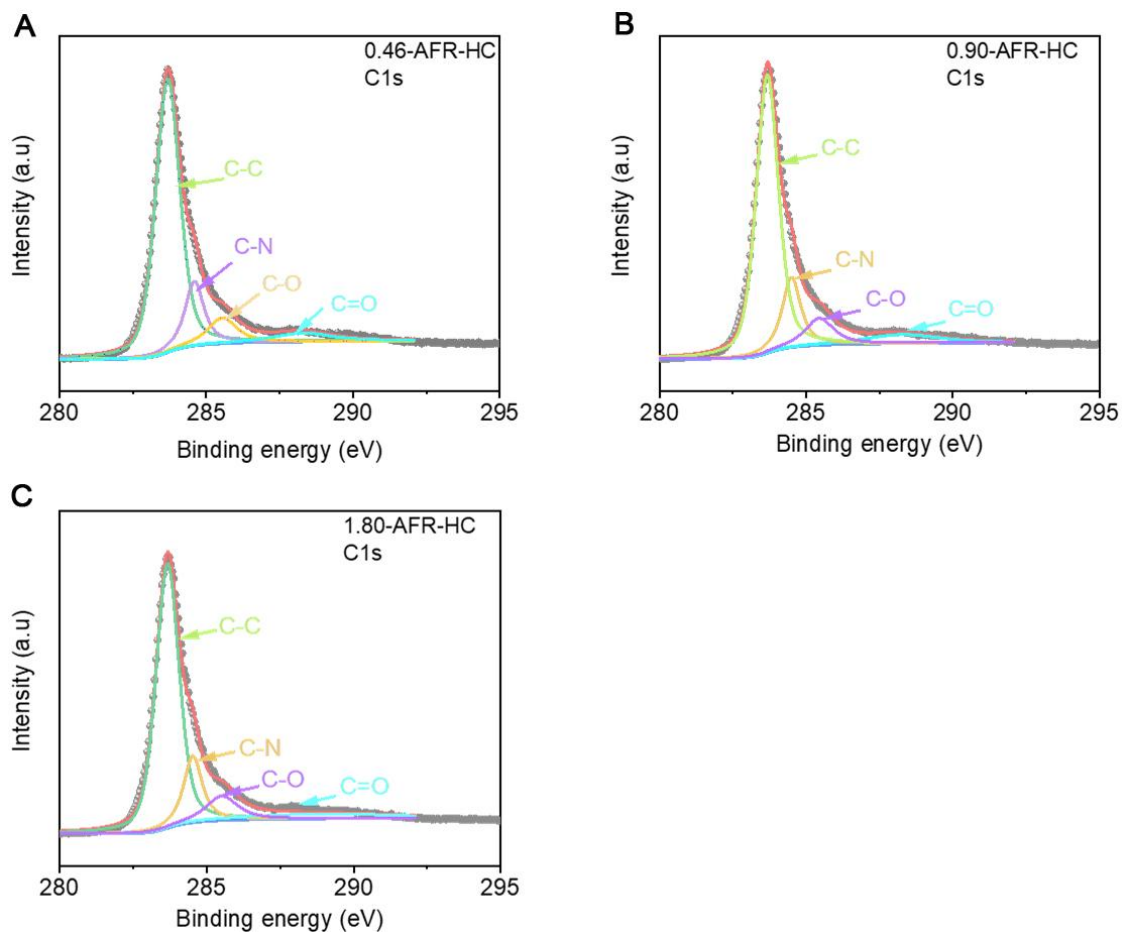
Supplementary Figure 3. XRD patterns. (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; and (C) 1.80-AFR-HC.



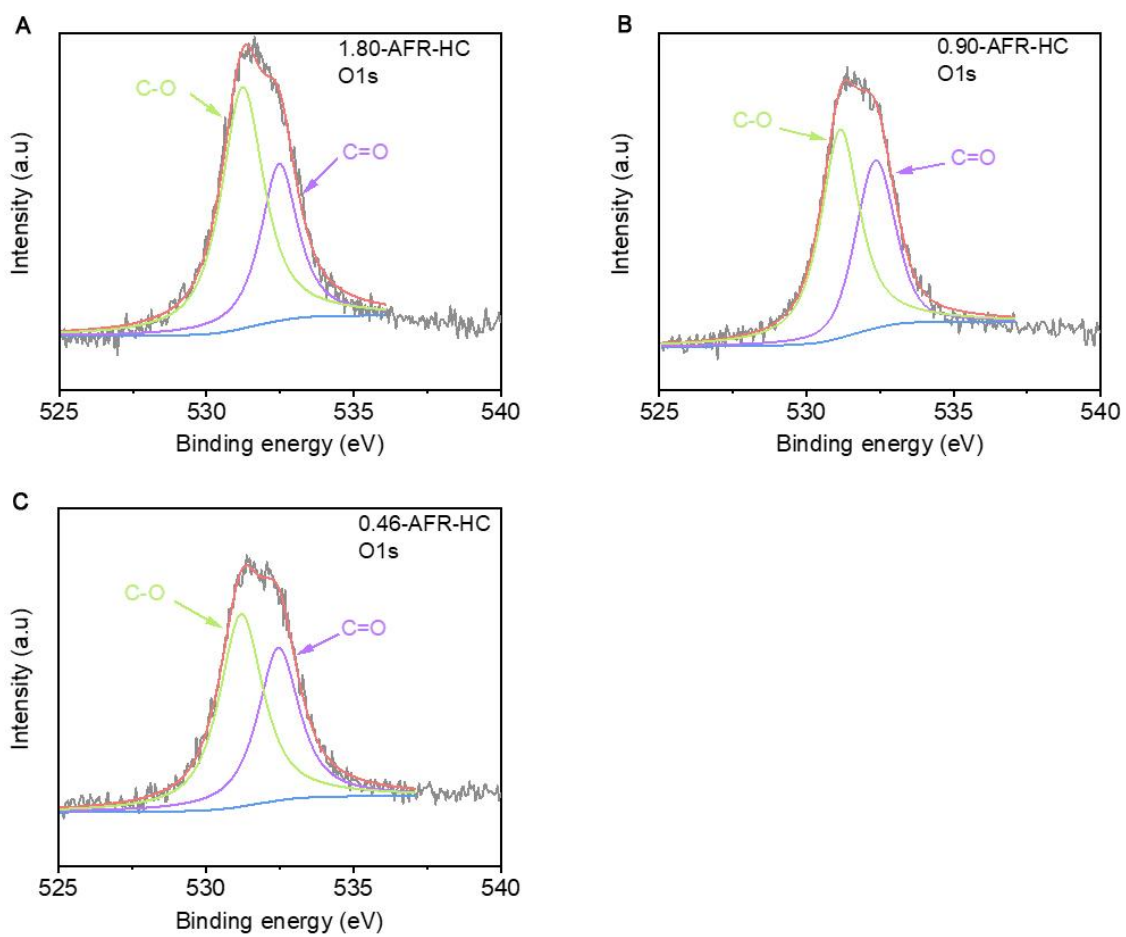
Supplementary Figure 4. Fitted Raman spectra. (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; and (C) 1.80-AFR-HC.



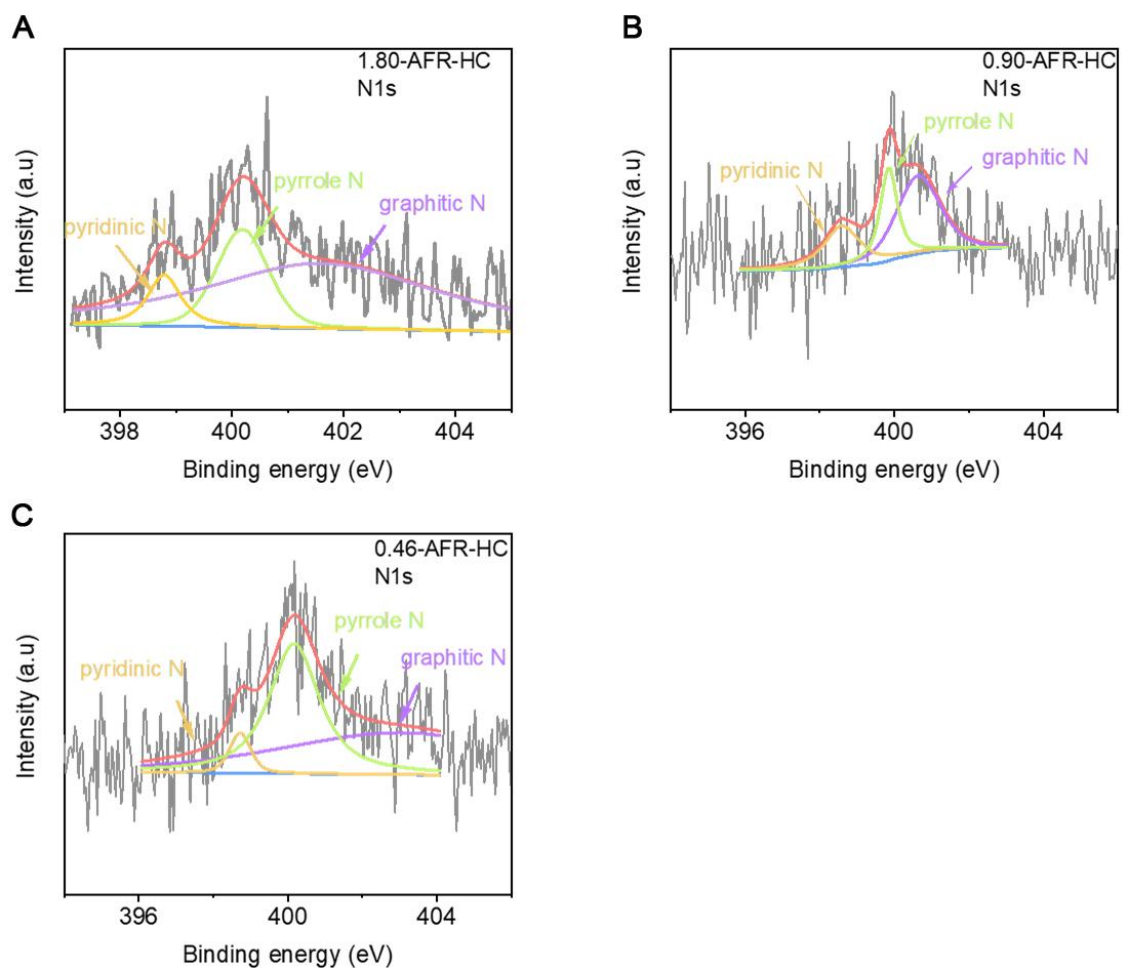
Supplementary Figure 5. The full XPS spectra. (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; (C) 1.80-AFR-HC.



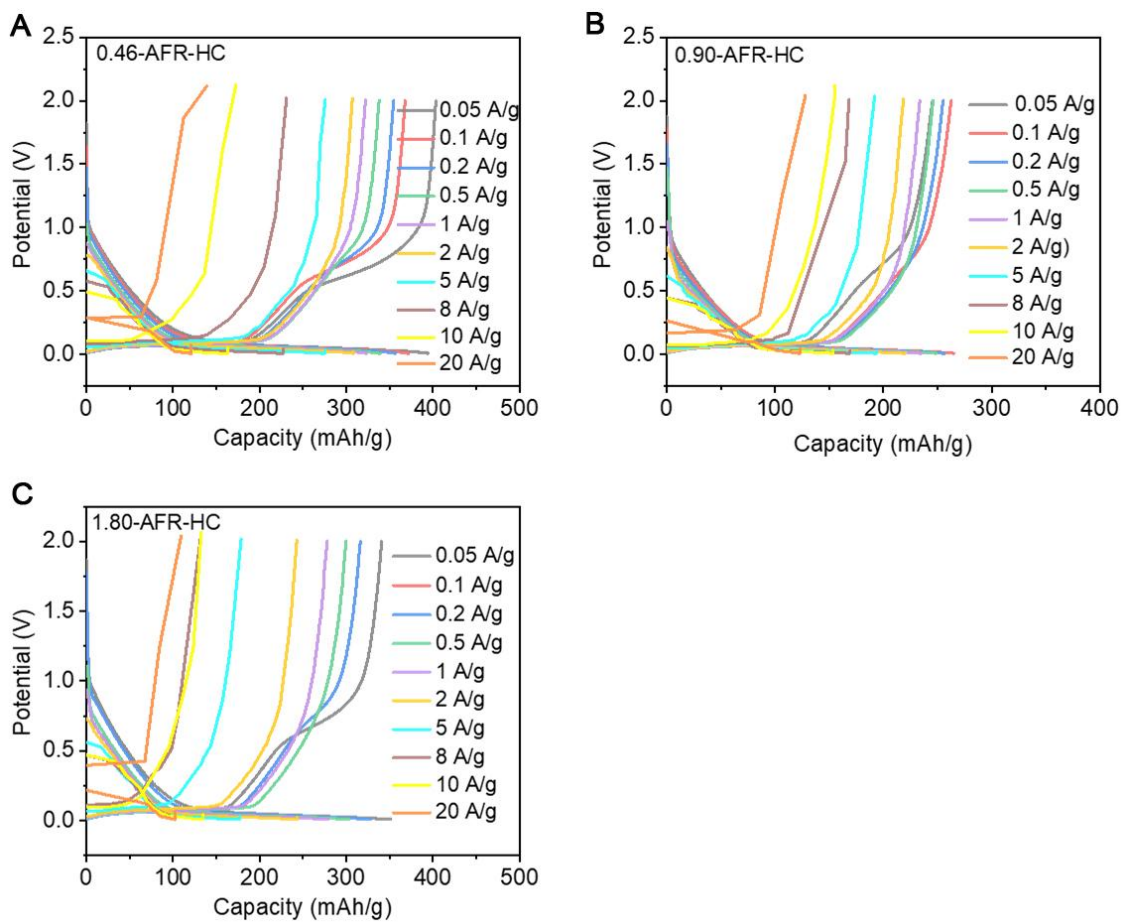
Supplementary Figure 6. The high-resolution XPS of C 1s spectra. (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; (C) 1.80-AFR-HC.



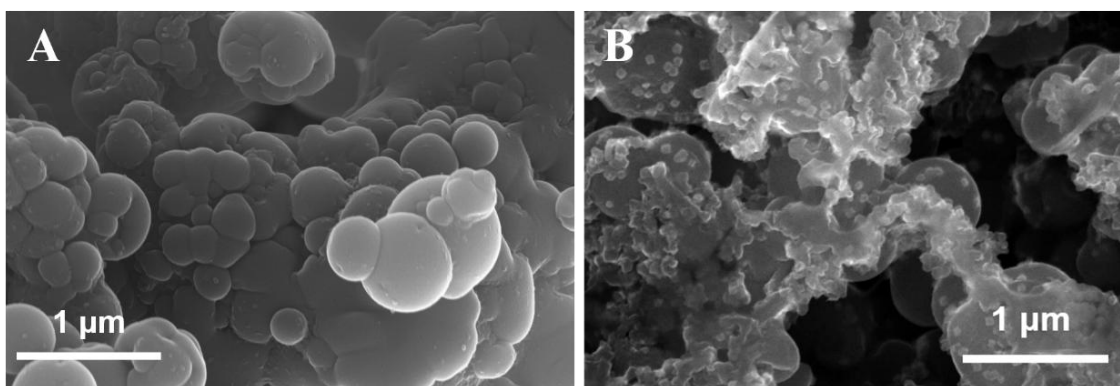
Supplementary Figure 7. The high-resolution XPS of O 1s spectra. (A) 1.80-AFR-HC; (B) 0.90-AFR-HC; (C) 0.46-AFR-HC.



Supplementary Figure 8. The high-resolution XPS of N 1s spectra. (A) 1.80-AFR-HC; (B) 0.90-AFR-HC; (C) 0.46-AFR-HC.

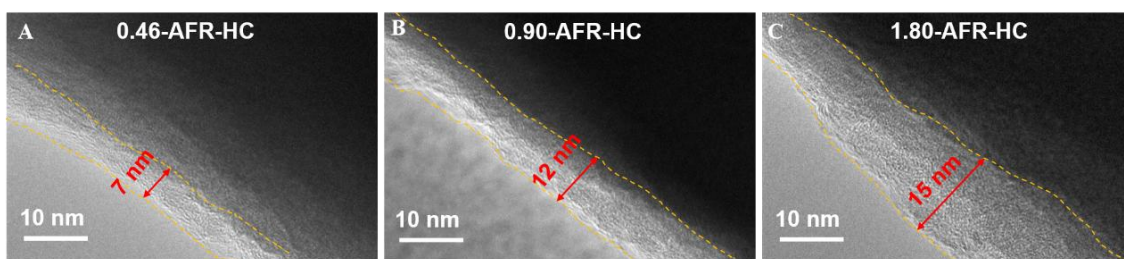


Supplementary Figure 9. The discharge-charge curves at different current densities (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; (C) 1.80-AFR-HC.

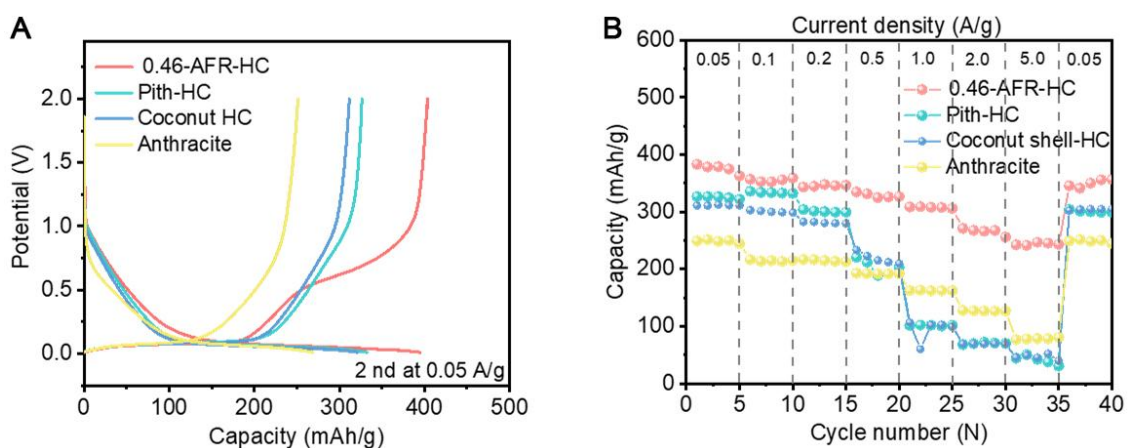


Supplementary Figure 10. (A) SEM images of 0.46-AFR-HC electrodes; (B) SEM images of 0.46-AFR-HC electrodes ($0.1 \text{ A} \cdot \text{g}^{-1}$, 1,000th cycle, $25 \text{ }^\circ\text{C}$).

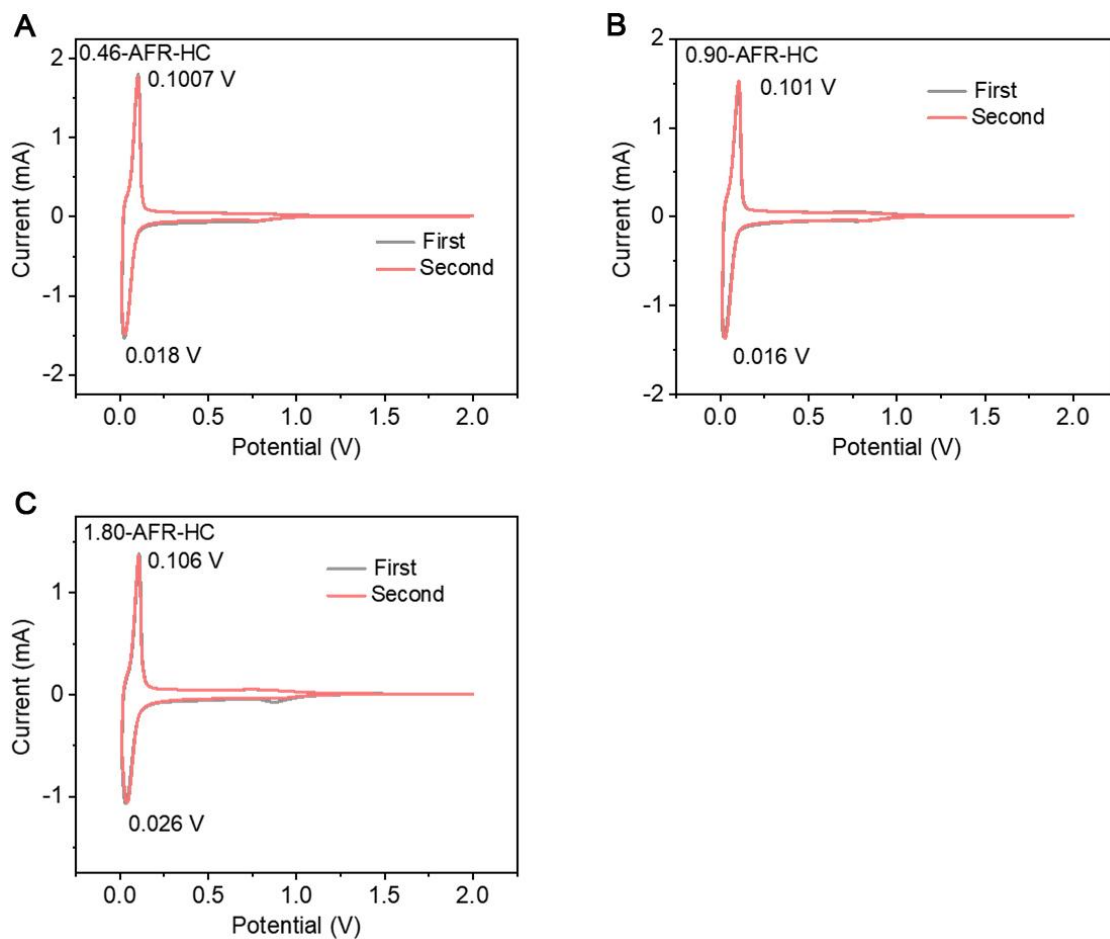
As shown in Supplementary Figure 10, the morphology of 0.46-AFR-HC can be maintained even after 1,000 cycles, demonstrating the good structural stability of 0.46-AFR-HC. Further HR-TEM images [Supplementary Figure 11] show that a 7-nm-thickness SEI layer is formed on the surface of 0.46-AFR-HC after 50 cycles, which is more uniform and thinner than that of 0.90-AFR-HC (12 nm) and 1.80-AFR-HC (15 nm). The uniform and thin SEI enable the faster interfacial Na^+ transportation, thus leading to higher ICE and rate capability.



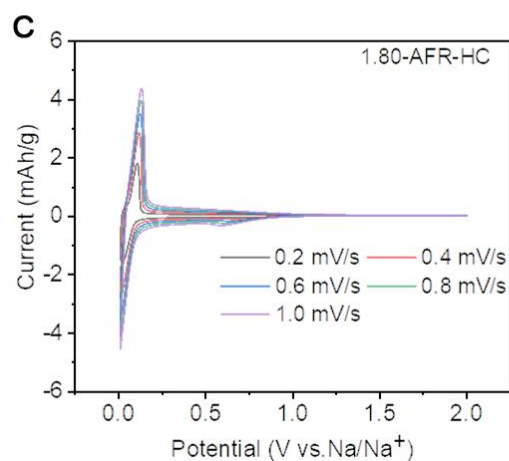
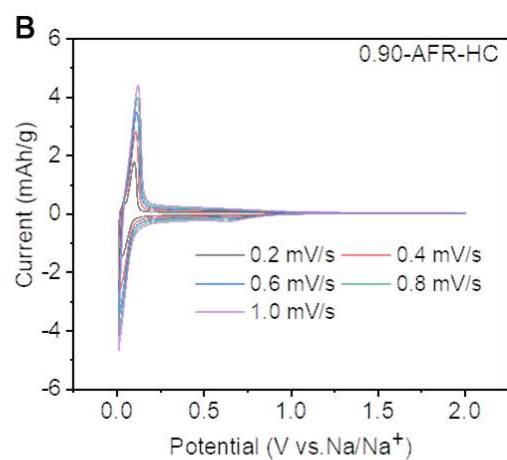
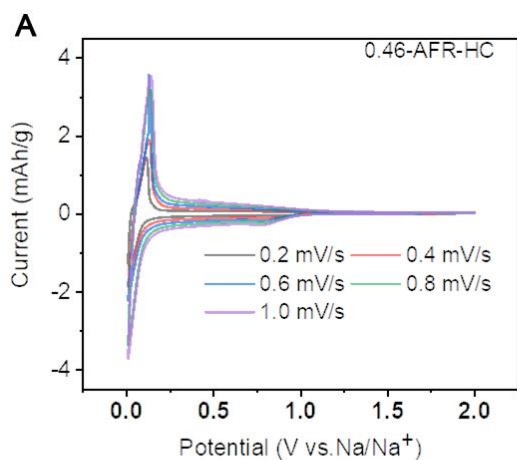
Supplementary Figure 11. HRTEM images of (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; and (C) 1.80-AFR-HC electrodes ($0.1 \text{ A}\cdot\text{g}^{-1}$, 50th cycle, 25°C).



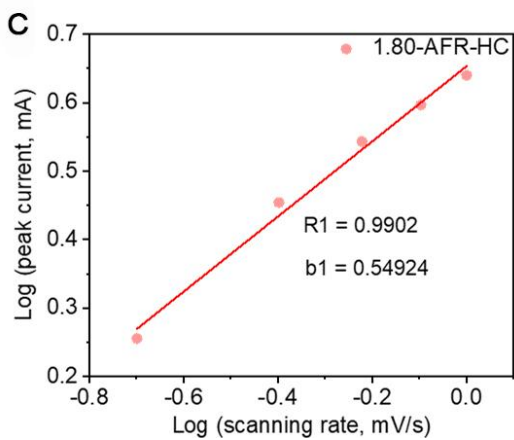
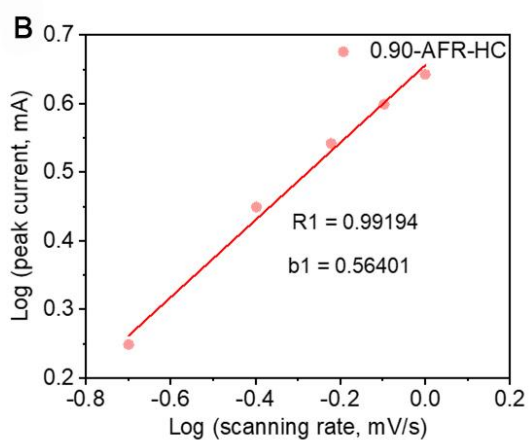
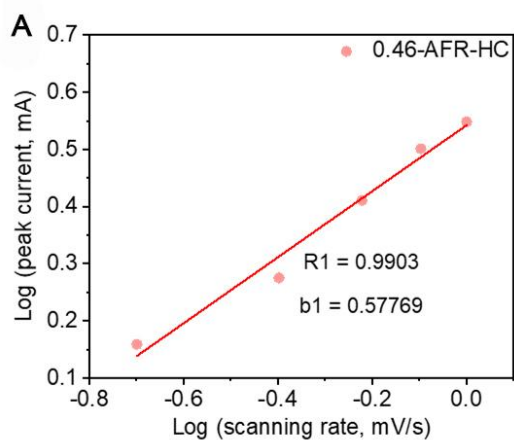
Supplementary Figure 12. (A) The GCD curves (the second cycle) of different hard carbons; (B) Rate performance of different types hard carbons.



Supplementary Figure 13. Cyclic voltammograms at scan rates of $0.2 \text{ mV}\cdot\text{s}^{-1}$. (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; (C) 1.80-AFR-HC.



Supplementary Figure 14. CV curves at different scan rates of 0.2-1.0 mV·s⁻¹. (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; (C) 1.80-AFR-HC.



Supplementary Figure 15. Fitting results of b value for (A) 0.46-AFR-HC; (B) 0.90-AFR-HC; and (C) 1.80-AFR-HC electrodes.

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