

Supplementary Materials

Axial chlorine-induced asymmetric cobalt single-atom coordination fields for boosting oxygen reduction reaction

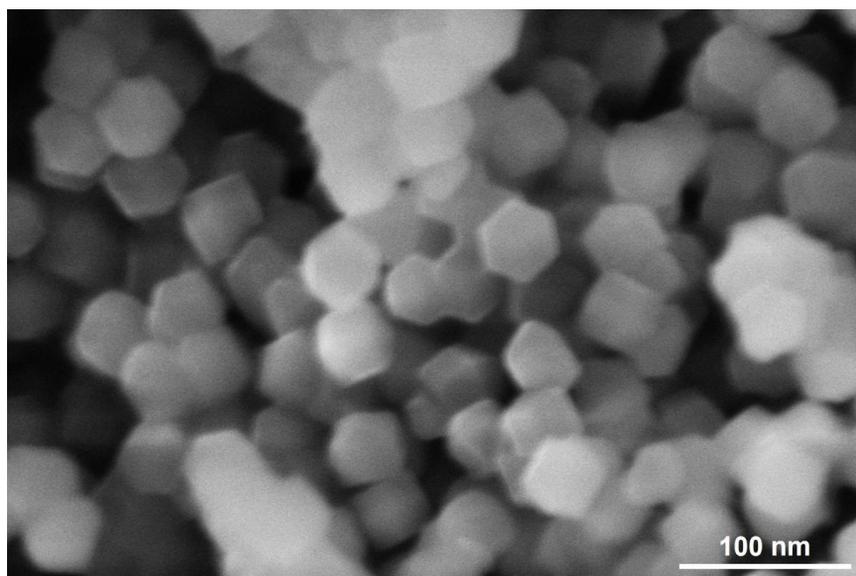
Xi-Rong Jiang¹, Guo-Dong Xie¹, Jun-Hao Li¹, Wen-Jie Huang¹, Jun-Da Lu¹, Pan Xie¹, Yan Dong¹, Wen-Da Ma¹, Yi-Da Deng^{1,2}, Xue-Rong Zheng^{1,2}

¹State Key Laboratory of Marine Resource Utilization in South China Sea, School of Materials Science and Engineering, Hainan University, Haikou 570228, Hainan, China.

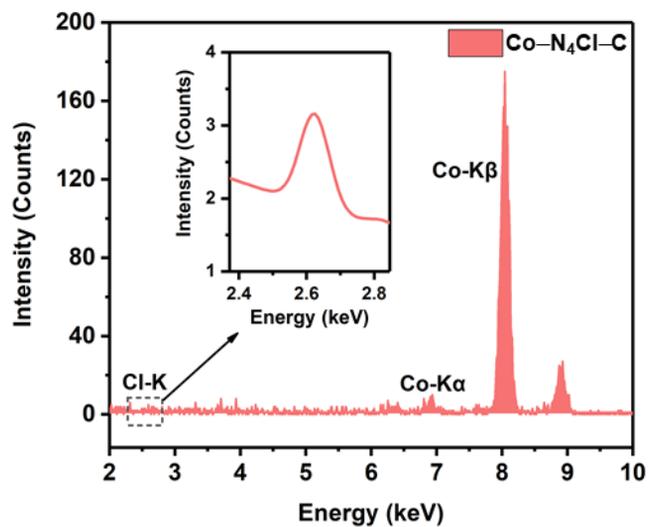
²School of Materials Science and Engineering, Key Laboratory of Advanced Ceramics and Machining Technology of Ministry of Education, Tianjin University, Tianjin 300072, China.

Correspondence to: Prof. Wen-Jie Huang, Prof. Xue-Rong Zheng, and Jun-Da Lu, State Key Laboratory of Marine Resource Utilization in South China Sea, School of Materials Science and Engineering, Hainan University, No. 58 Renmin Avenue, Haikou 570228, Hainan, China. E-mail: wj_huang@hainanu.edu.cn; xrzh@hainanu.edu.cn; lujundaacc@163.com

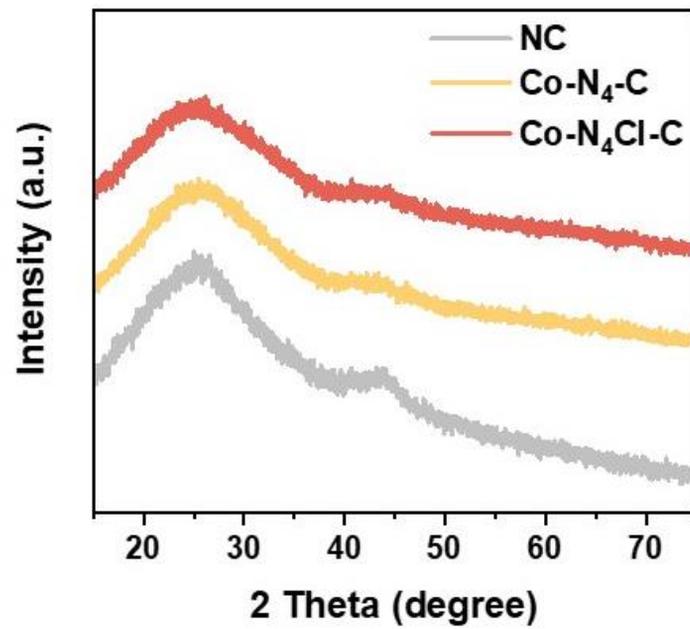
Supplementary figures



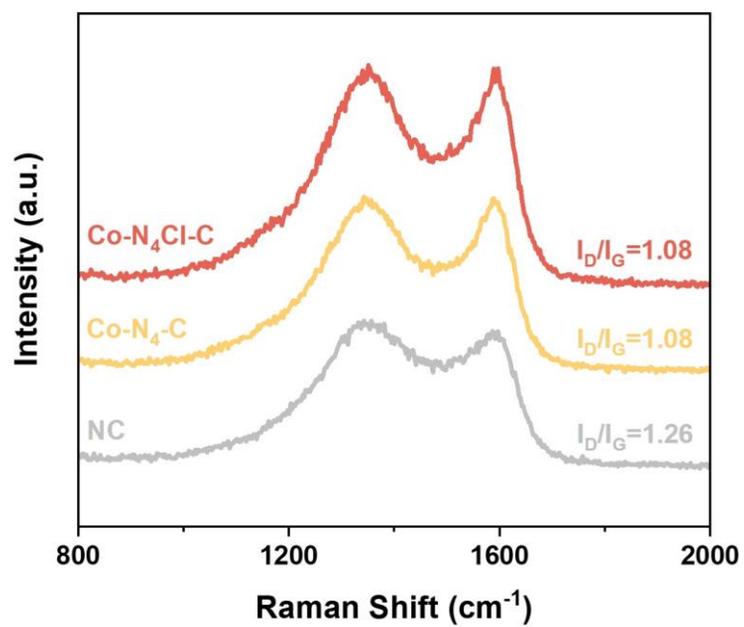
Supplementary Figure 1. SEM image of Co-ZIF-8 precursor.



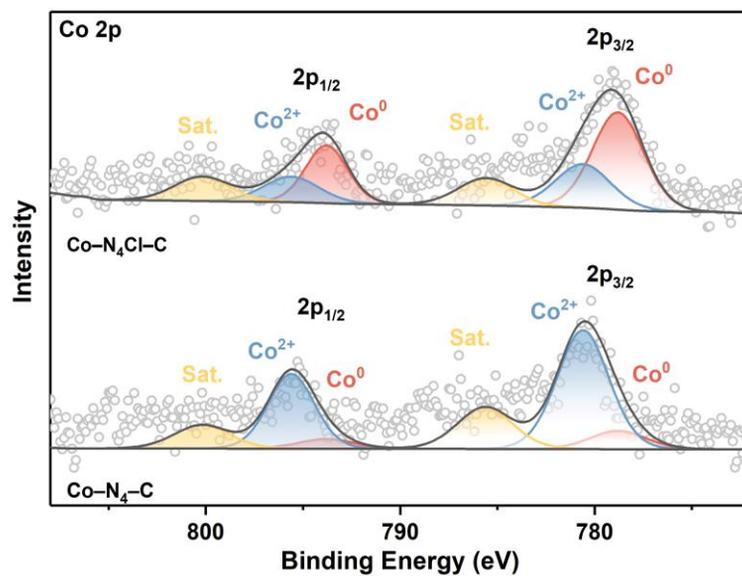
Supplementary Figure 2. The fitting curves of the energy dispersive spectroscopy spectrum of Co-N₄Cl-C catalyst.



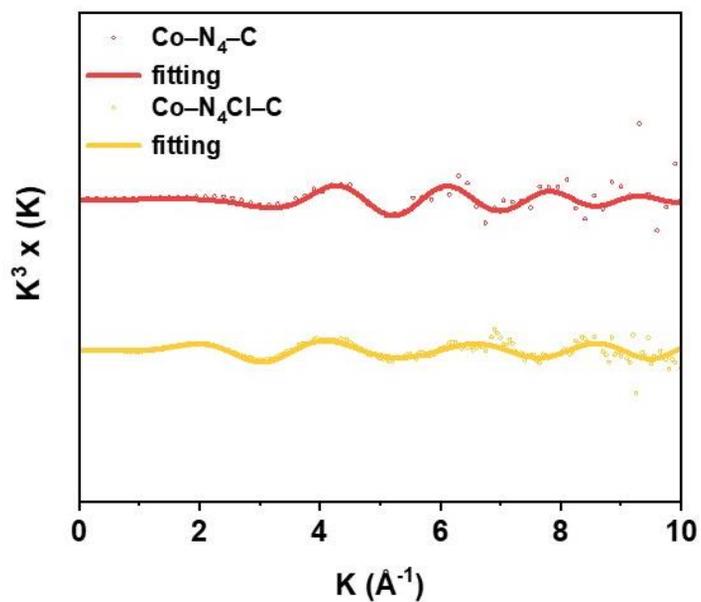
Supplementary Figure 3. XRD patterns of Co-N₄-C, Co-N₄Cl-C and NC catalysts.



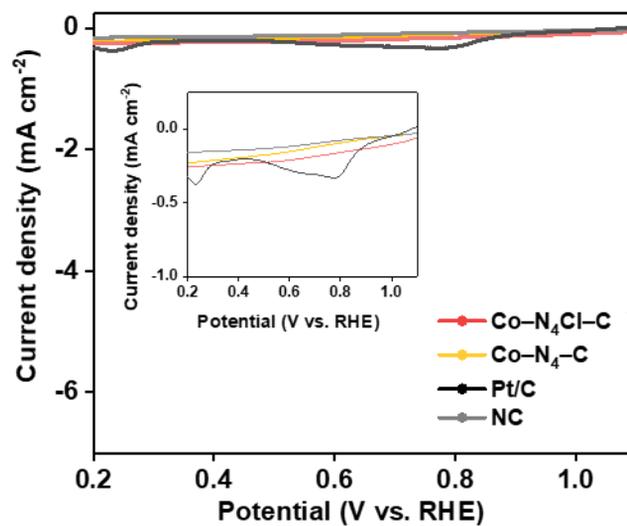
Supplementary Figure 4. Raman spectra of Co-N₄-C and Co-N₄Cl-C catalysts.



Supplementary Figure 5. Co 2p XPS spectra of Co-N₄-C and Co-N₄Cl-C catalysts.



Supplementary Figure 6. Extended X-ray absorption fine structure (EXAFS) fitting curves of the K-edge of Co-N₄-C and Co-N₄Cl-C catalysts at the *k*-space.

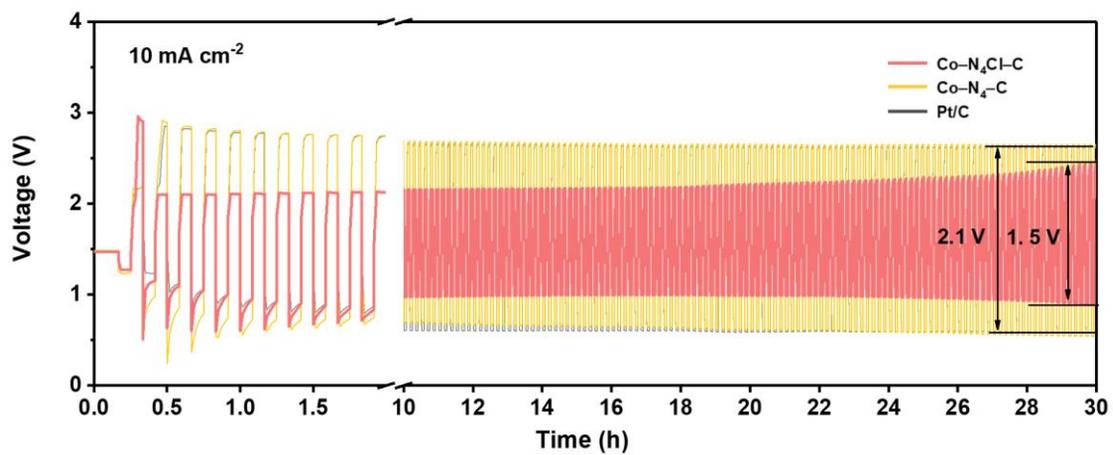


Supplementary Figure 7. Oxygen reduction reaction polarization curves of Co-N₄-C, Co-N₄Cl-C, Pt/C, and NC at a scan rate of 5 mV·s⁻¹ under a rotation speed of 1,600 rpm in N₂-saturated 0.5 M NaCl electrolyte containing 1 M KOH.



Supplementary Figure 8. Photographic images of the open circuit voltage for (A) Co-N₄Cl-C-, (B) Co-N₄-C-, and (C) Pt/C-based SWZABs.

The Co-N₄Cl-C-based seawater-based zinc-air battery (SWZAB) exhibits an open-circuit voltage of 1.460 V, surpassing that of SWZABs driven by Co-N₄-C (1.345 V) and Pt/C (1.354 V). These values are in good agreement with the results measured using a workstation.



Supplementary Figure 9. Cycling performance of Co-N₄Cl-C-, Co-N₄-C- and Pt/C-based SWZABs at 10 mA·cm⁻².

Supplementary tables

Supplementary Table 1. Comparison of inductively coupled plasma OES spectrometer (ICP-MS) analysis for Co and Zn element in Co-N₄-C and Co-N₄Cl-C

Characterization	ICP-MS (wt.%)for Zn	ICP-MS (wt.%)for Co
Co-N ₄ -C	0.002	1.27
Co-N ₄ Cl-C	0.002	1.44

Trace amounts of Zn in the ICP-MS analysis indicate that Zn is effectively removed in Co-N₄-C and Co-N₄Cl-C.

Supplementary Table 2. Structural parameters extracted from quantitative EXAFS curve-fitting of samples using the ARTEMIS module of IFEFFIT

Sample	Path	<i>N</i>	<i>R</i> (Å)	σ^2 (10^{-3}Å^2)	ΔE_0 (eV)	R-factor
Co-N ₄ Cl-C	Co-N	3.6±0.5	2.03±0.01	1.58±0.04	-2.552±0.7	0.008
	Co-Cl	0.9±0.2	2.26±0.01	5.40±0.06		
Co-N ₄ -C	Co-N	4.1±0.3	1.88±0.01	9.84±0.04	-1.31±0.2	0.008

[a] *N*, coordination number; [b] *R*, distance between absorber and backscatter atoms; [c] σ^2 , Debye-Waller factor to account for both thermal and structural disorders; [d] ΔE_0 , inner potential correction; [e] **R-factor** (%) indicates the goodness of the fit.