

1 **Supplementary Material:**

2 **Regulating the Electrocatalytic Performance for Nitrogen**
3 **Reduction Reaction by Tuning the N Contents in Fe₃@N_xC_{20-x} (x =**
4 **0~4): A DFT Exploration**

5 **How to Use This Template**

6 *This template shows the manuscript structure that can be used in supplementary material. Please note that each*
7 *part has a corresponding style, which authors should follow. Please note that the fonts in gray show writing*
8 *requirements. For any questions, you may contact the editorial office.*

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15 **Supplementary Table 1. The corrected values of gas molecules (in eV).**

Gas	
N ₂	-0.35
H ₂	-0.05
NH ₃	0.42

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23 Supplementary Table 2. The corrected values of intermediates on Fe₃@C₂₀ and Fe₃@N₄C₁₆ (in eV).

Reaction routes		Intermediates						
Fe ₃ @C ₂₀	Consecutive	*NN	*NNH	*NNH ₂	*N	*NH	*NH ₂	*NH ₃
		0.13	0.44	0.65	0.08	0.34	0.67	0.95
	Enzymatic	*NN	*NNH	*NHNH	*NH ₂ NH	*NH ₂ NH ₂	*NH ₂	*NH ₃
		0.13	0.44	0.74	1.08	1.42	0.67	0.95
	Distal	*NN	*NNH	*NNH ₂	*N	*NH	*NH ₂	*NH ₃
		0.10	0.43	0.73	0.07	0.34	0.66	0.95
Alternative	*NN	*NNH	*NHNH	*NH ₂ NH	*NH ₂ NH ₂	*NH ₂	*NH ₃	
	0.10	0.43	0.78	0.97	1.42	0.66	0.96	
Fe ₃ @N ₄ C ₁₆	Consecutive	*NN	*NNH	*NNH ₂	*N	*NH	*NH ₂	*NH ₃
		0.08	0.44	0.65	0.08	0.34	0.67	0.95
	Enzymatic	*NN	*NNH	*NHNH	*NH ₂ NH	*NH ₂ NH ₂	*NH ₂	*NH ₃
		0.08	0.44	0.74	1.08	1.42	0.67	0.95

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28 Supplementary Table 3. Formation energies (E_f, in eV) of Fe₃@N_xC_{20-x} with Fe₂-bonded C replaced**29 by N (x is the number of N atoms).**

x	1	2-I	2-II	2-III	3	4
structure	Supplementary Figure 4A	Supplementary Figure 4B	Supplementary Figure 4C	Supplementary Figure 4D	Supplementary Figure 4E	Supplementary Figure 4F
E _f	-0.20	-0.22	-0.50	-0.35	-0.33	-0.65

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40 **Supplementary Table 4. Lattice constants (a and b , in Å) of $\text{Fe}_3\text{@N}_x\text{C}_{20-x}$, nitrogen adsorption**
 41 **energies on Fe_2 sites and Fe_1 sites (E_{ads} , in eV).**

	a	b	Fe2 site		Fe1 site	
			E_{ads} (end-on)	E_{ads} (side-on)	E_{ads} (end-on)	E_{ads} (side-on)
Fe₃@C₂₀	11.14	7.95	-0.52	-0.57	-0.41	-0.10
Fe₃@N₁C₁₉	11.08	7.90	-0.55	-0.62	-0.43	-0.11
Fe₃@N₂C₁₈-I	11.00	7.91	-0.59	-0.55	-0.39	-0.06
Fe₃@N₂C₁₈-II	11.02	7.73	-0.50	-0.48	-0.47	-0.08
Fe₃@N₂C₁₈-III	11.04	7.66	-0.59	-0.52	-0.43	-0.17
Fe₃@N₃C₁₇	10.95	7.85	-0.54	-0.52	-0.58	-0.08
Fe₃@N₄C₁₆	10.87	7.74	-0.32	-0.20	-0.44	0.15

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44 **Supplementary Table 5. The free energy change (ΔG , in eV) of the step $^*\text{NN} \rightarrow ^*\text{NNH}$ (in eV).**

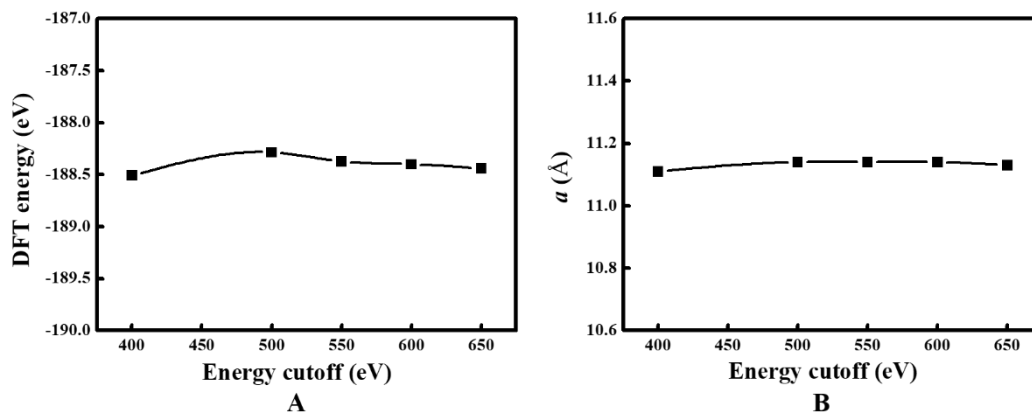
	end-on	side-on
Fe₃@C₂₀	0.59	0.64
Fe₃@N₁C₁₉	1.14	0.74
Fe₃@N₂C₁₈-I	1.09	0.61
Fe₃@N₂C₁₈-II	1.07	0.73
Fe₃@N₂C₁₈-III	0.65	0.71
Fe₃@N₃C₁₇	1.08	0.67
Fe₃@N₄C₁₆	/	0.45

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46 **Supplementary Table 6. The Fe–Fe bond lengths ($d_{\text{Fe-Fe}}$) and d -band centers and p -band centers of**
 47 **the Fe₂-bonded C/N for $\text{Fe}_3\text{@N}_x\text{C}_{20-x}$ ($x = 0\sim 4$).**

	$d_{\text{Fe-Fe}}$ (Å)	d -band centers	p -band centers of Fe ₂ -bonded C and N atoms
Fe₃@C₂₀	2.22	-0.92	-3.51
Fe₃@N₁C₁₉	2.22	-0.92	-3.57
Fe₃@N₂C₁₈-I	2.23	-0.91	-4.43
Fe₃@N₂C₁₈-II	2.22	-1.05	-4.02
Fe₃@N₂C₁₈-III	2.24	-1.05	-3.96
Fe₃@N₃C₁₇	2.23	-1.00	-4.81
Fe₃@N₄C₁₆	2.16	-0.98	-3.89

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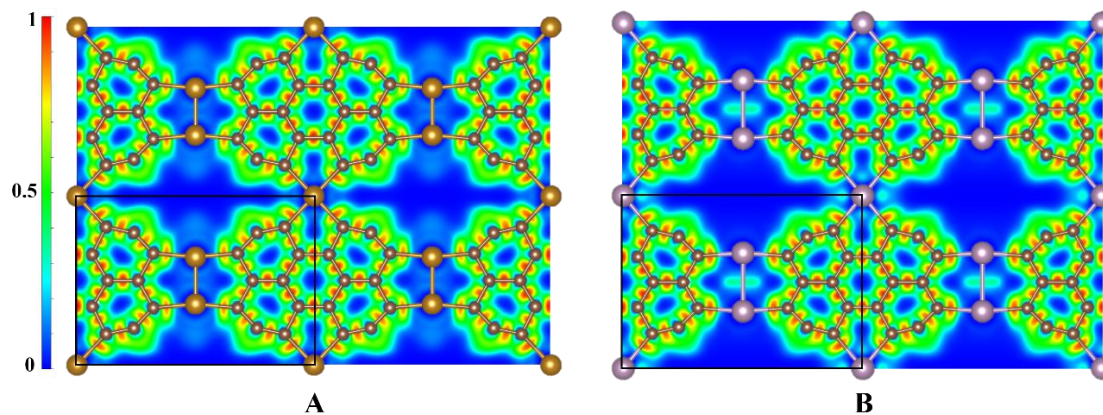
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50 **Supplementary Figure 1.** Evaluation of energy cutoff for optimizing Fe₃@C₂₀: the variation of the total
 51 DFT energy (A) and the lattice constant *a* (B) against energy cutoff.

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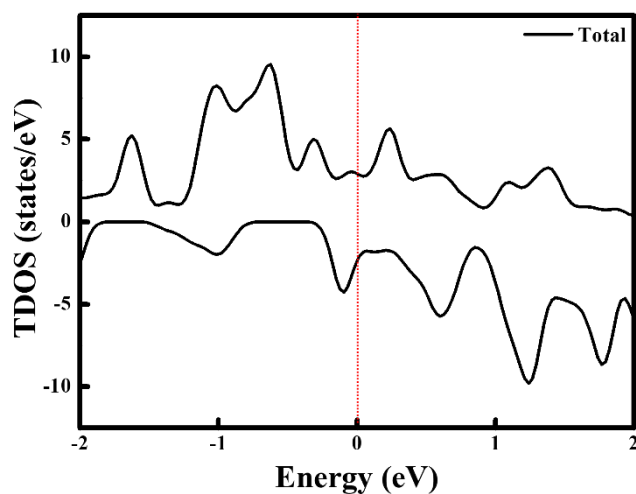
57 **Supplementary Figure 2.** The ELF plot of (110) section of Fe₃@C₂₀(A) and Mo₃@C₂₀(B) monolayer,
 58 with the unit cells marked by black lines.

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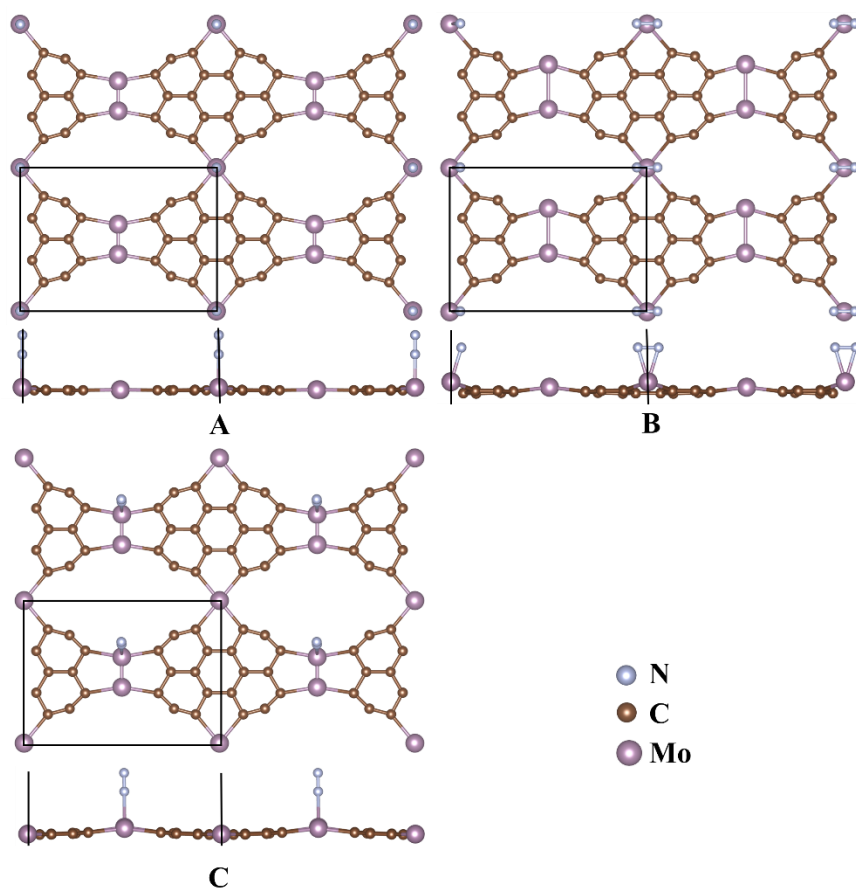


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64 **Supplementary Figure 3.** The TDOS of 2D Mo₃@C₂₀ monolayer with the Fermi energy level indicated
 65 by the red dashed line.

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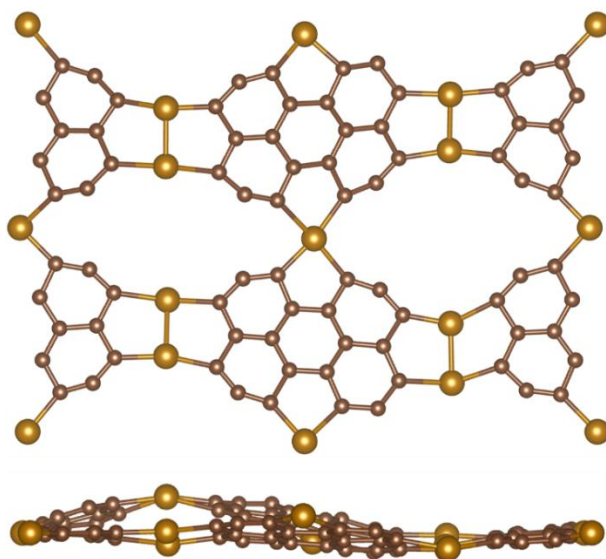
69 **Supplementary Figure 4.** Different adsorption structures of N₂ on Mo₃@C₂₀: (A) end-on and (B) side-on
 70 configuration at Mo₁ site, and (C) end-on and configuration at Mo₂ site, with the unit cells marked by black
 71 lines.

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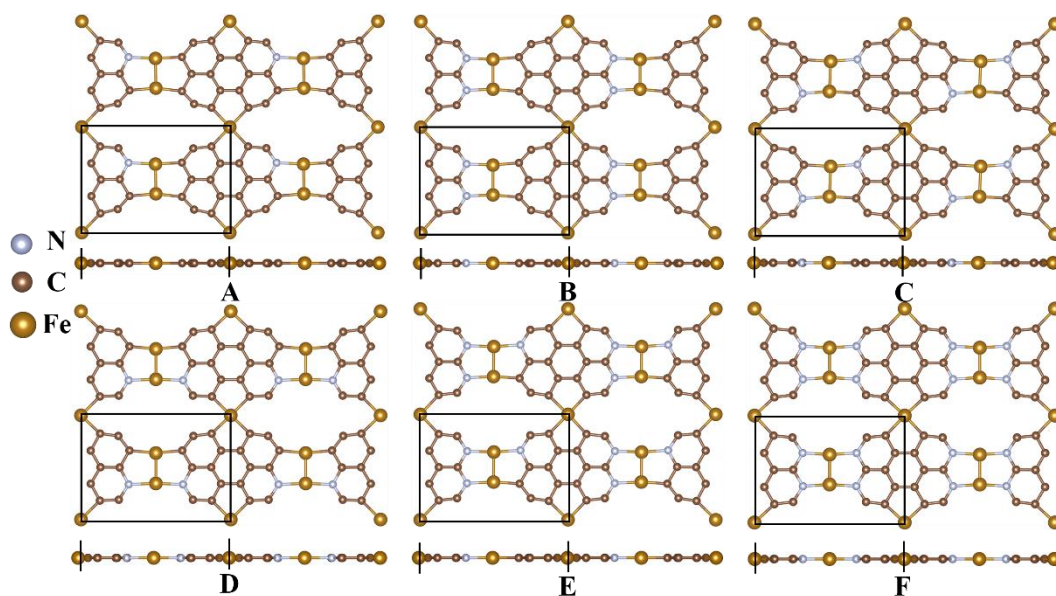
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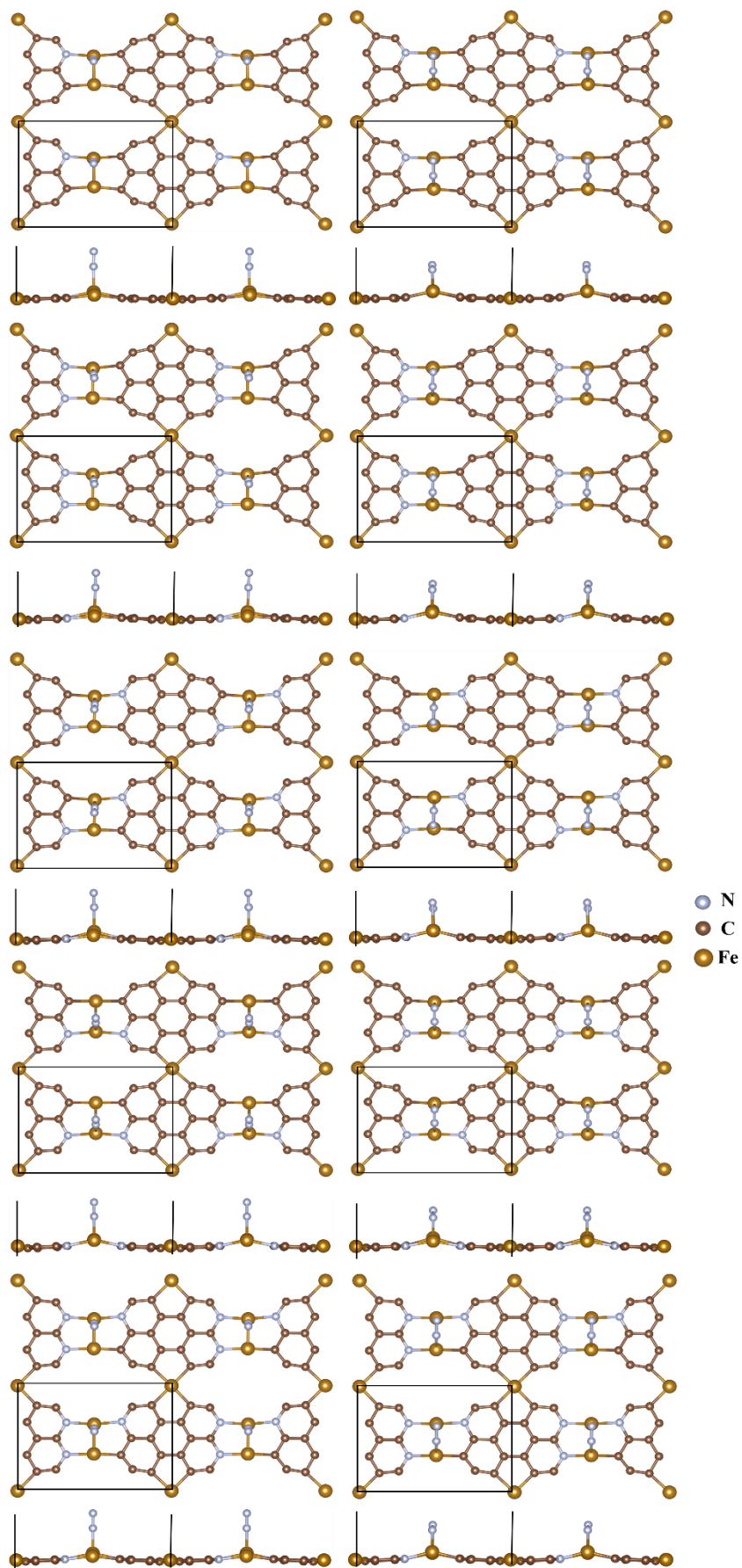
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Supplementary Figure 5. The final structure of $\text{Fe}_3@C_{20}$ after 5-ps' FPMD simulations at 300 K.



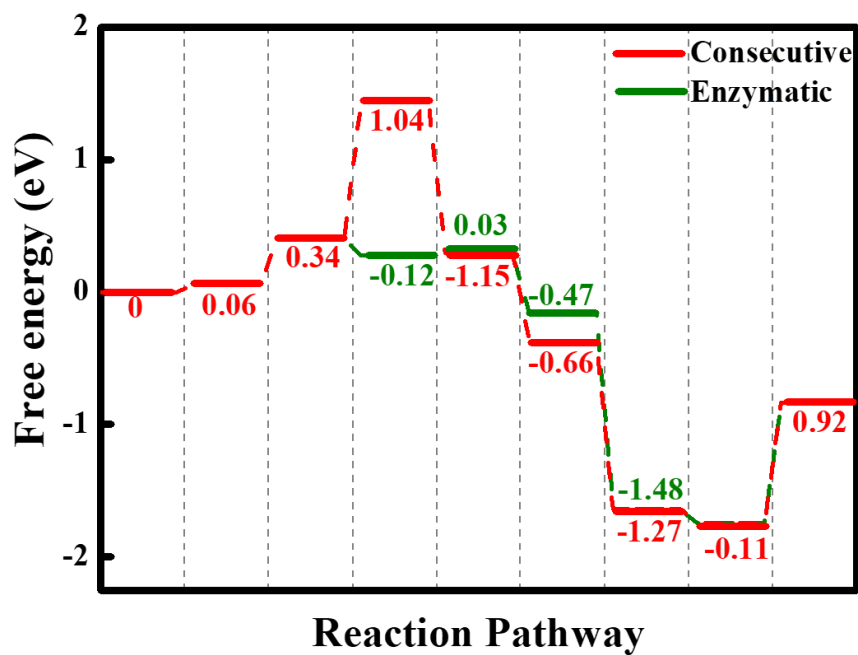
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Supplementary Figure 6. The top and side views of $\text{Fe}_3@N_1C_{19}$ (A), $\text{Fe}_3@N_2C_{18}$ -I (B), $\text{Fe}_3@N_2C_{18}$ -II (C), $\text{Fe}_3@N_2C_{18}$ -III (D), $\text{Fe}_3@N_3C_{17}$ (E), and $\text{Fe}_3@N_4C_{16}$ (F), with the unit cells marked by black lines.



Supplementary Figure 7. Different adsorption structures of N_2 adsorption on $Fe_3@N_xC_{20-x}$ ($x = 1\sim 3$), with the unit cells marked by black lines.

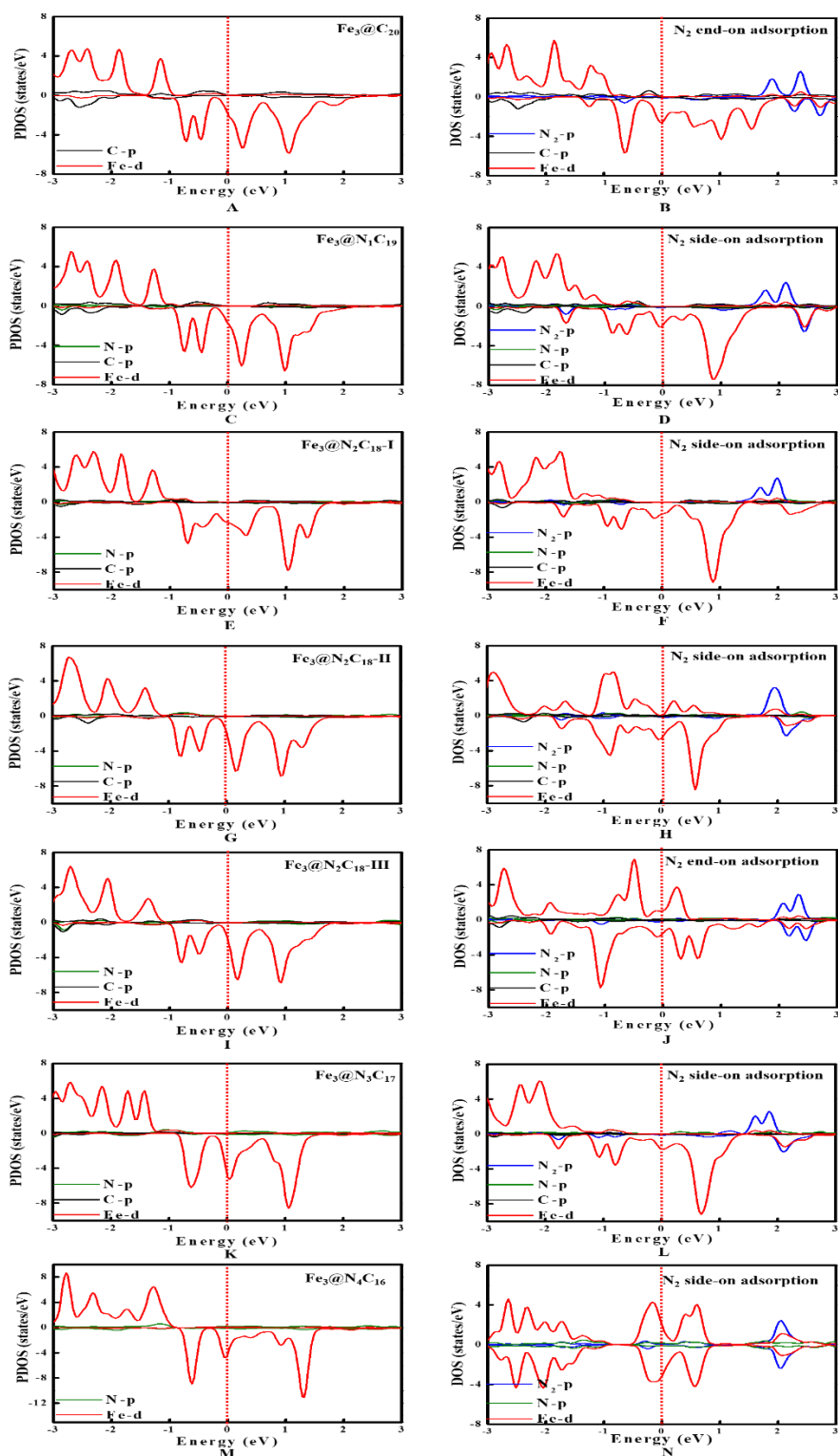
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91 **Supplementary Figure 8.** The free energy diagram of eNRR through consecutive/ enzymatic (in red/green)
 92 pathway on $\text{Fe}_3\text{@N}_4\text{C}_{16}$ including the solvation effect. Data denote the ΔG of each elementary step.

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95 **Supplementary Figure 9.** The PDOS of $\text{Fe}_3@N_x\text{C}_{20-x}$ ($x = 0\sim 4$) structures without and with N_2 adsorption
 96 for $\text{Fe}_3@C_{20}$ (A,B) $\text{Fe}_3@N_1C_{19}$ (C,D), $\text{Fe}_3@N_2C_{18}$ -I (E,F), $\text{Fe}_3@N_2C_{18}$ -II (G,H), $\text{Fe}_3@N_2C_{18}$ -III (I,J),
 97 $\text{Fe}_3@N_3C_{17}$ (K,L), and $\text{Fe}_3@N_4C_{16}$ (M,N). Only the d -orbital of the Fe_2 site and its 4 bonded C/N atoms
 98 are considered in the figure. The black, red, green and blue solid lines represent the C- p , Fe- d , doped N- p
 99 and $^*\text{N}_2$ molecular p orbitals, respectively, and the red dashed line indicates the Fermi energy level.