## **Supplementary Material**

Composition-regulated lattice strain of PdSn/C for boosting C1 pathway in ethanol electrooxidation

Yin Cai<sup>1,#</sup>, Yi Tao<sup>1,#</sup>, Jie Ding<sup>2</sup>, Fuhua Li<sup>3</sup>, Rongsheng Chen<sup>1</sup>, Tao Ma<sup>1,\*</sup>, Feng Liang<sup>1,\*</sup>

<sup>1</sup>The State Key Laboratory of Refractories and Metallurgy, School of Chemistry and Chemical Engineering, Wuhan University of Science and Technology, Wuhan 430081, Hubei, China.

<sup>2</sup>Department of Materials Science and Engineering, City University of Hong Kong, Hong Kong, SAR 999077, China.

<sup>3</sup>School of Chemistry and Chemical Engineering, Chongqing Key Laboratory of Theoretical and Computational Chemistry, Chongqing University, Chongqing 401331, China.

<sup>#</sup>Authors contributed equally.

**Correspondence to:** Dr. Tao Ma, The State Key Laboratory of Refractories and Metallurgy, School of Chemistry and Chemical Engineering, Wuhan University of Science and Technology, No. 947, Heping Avenue, Qingshan District, Wuhan 430081, Hubei, China. E-mail: <u>taoma01@wust.edu.cn</u>; Prof. Feng Liang, The State Key Laboratory of Refractories and Metallurgy, School of Chemistry and Chemical Engineering, Wuhan University of Science and Technology, No. 947, Heping Avenue, Qingshan District, Wuhan 430081, Hubei, China. E-mail: <u>feng\_liang@whu.edu.cn</u>



**Figure S1.** High-magnification TEM images (left) and size distribution (right) of (A) Pd/C, (B) PdSn/C-0.1, (C) PdSn/C-0.5, (D) PdSn/C-1, (E) PdSn/C-2. The scale in TEM image is 5 nm.



**Figure S2.** TEM images and corresponding EDS mappings of (A) PdSn/C-0.1, (B) PdSn/C-0.5, (C) PdSn/C -1, and (D) PdSn/C-2.



**Figure S3.** (A) The density of states (DOS) for Pd alloy with different tensile effect, (B) the change of d-band center due to strain of the Pd (111) facet.



**Figure S4.** (A) Wide scan XPS spectra of PdSn/C-X, and its high-resolution region of (B) Pd 3d, (C) Sn 3d, and (D) O 1s.



**Figure S5.** (A) Summarized ECSAs of different PdSn/C-X and Pd/C in 1.0 M KOH, (B) LSV curves, (C) Tafel plots, and (D) Nyquist plots of different PdSn/C-X and Pd/C collected in 1.0 M KOH and 1.0 M ethanol. The insets in (B) and (D) are the magnified view around onset potential and the equivalent circuit model used to fit the Nyquist plots, respectively.



**Figure S6.** (A) The CVs before and after 1500 sweeps of PdSn/C-1 recorded in 1.0 M KOH with 1.0 M ethanol solution at scan rate of 50 mV/s. (B) The plots of forward peak current density of PdSn/C-1 and Pd/C based on 1500 sweeps. (C) The PdSn/C-1 loaded GCE before and after 1500 sweeps.



**Figure S7.** (A) TEM image, (B) size distribution, (C) corresponding EDS mappings, and (D) HAADF-STEM image of PdSn/C-1 after 1500 sweeps.



**Figure S8.** CO stripping voltammograms of PdSn/C-X and Pd/C obtained in 1.0 KOH.



**Figure S9.** (A) The fitted bands (corresponding to Figure 4B) at 1405 to 1391 cm<sup>-1</sup> of PdSn/C-1, (B) the corresponding integrated absorbances of the peaks around 1394 and 1403 cm<sup>-1</sup> (left axis) and selectivity to C1 (right axis) for PdSn/C-1.



**Figure S10.** The atomic structures of reaction intermediates during EOR on the Pd (111) surface.

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Catalysts	Pd (mg/L)	Sn (mg/L)	C (mg/L)	Pd (%)	Sn (%)	Sn/Pd (mol)
PdSn/C-0.1	431.53	34.50	2491.68	14.59	1.17	1:14
PdSn/C -0.5	510.44	46.32	3112.83	13.91	1.26	1:12
PdSn/C -1	399.09	94.951	3995.16	8.89	2.12	1:5
PdSn/C -2	428.56	109.94	2520.46	14.01	3.59	1:4
Pd/C	456.82	0	2227.28	17.02	0	0:1

Table S1. Element content of as-prepared catalysts according to ICP-MS analysis

Table S2. Comparison of the electrochemical performance of various nanomaterials for
EOR in alkaline media

Catalyst	Method	Electrolyte	Mass activity (mA/mg <sub>Pd</sub> )	ECSA (m <sup>2</sup> /g <sub>Pd</sub> )	Stability	Ref.
PdAg	Hydrothermal	1.0 M KOH and 1.0 M ethanol	1950	24.2	342 mA mg <sup>-1</sup> after 1000 s.	[4]
Pd/NiMoO4	Hydrothermal	1.0 M KOH and 1.0 M ethanol	3250	174.2	10.62 mA cm <sup>-2</sup> after 10,000 s	[5]
PdSn:P/C	One-pot	0.5 M KOH and 0.5 M ethanol	5030	120.1	9.52 mA cm <sup>-2</sup> after 7000s	[6]
PdAg@Pd	One-pot	1.0 M KOH and 1.0 M ethanol	3480	28.11	190.6 mA mg <sub>Pd</sub> after 3600s	[7]
PdAuPt	Surfactant-free method	1.0 M KOH and 1.0 M	4990	62.92	92% After 500 cycles	[8]

		ethanol				
PdBP	One-sten	1.0 M KOH			1870 mA	
	aqueous	and 1.0 M	4150	35	$mg_{Pd}$ after	[9]
	aqueous	ethanol			5000 cycles	
PdPb	One-step hydrothermal	1.0 M KOH			1497 mA	
		and 1.0 M	3460	62	mg <sub>Pd</sub> after	[10]
		ethanol			1000s	
		1.0 M KOH		12	47.20/ - 8	[11]
PdPtBi	Hydrothermal	and 1.0 M	11080		47.2% after	
		ethanol			3,600 s	
	Hydrothermal	1.0 M KOH		21.8	152 mA mg <sub>Pd</sub> after 6000s	[12]
PtAg		and 1.0 M	6100			
		ethanol				
		1.0 M KOH			289mA mgPd	
AgPdPt	One-pot	and 1.0 M	7240	103.8	after for 7200	[13]
		ethanol			s	
	Co-reduction	1.0 M NaOH			decay only	
PdSn/TiO <sub>2</sub>		and 1.0M	3381	NA	20% after 300	[14]
		ethanol			cycles	
	Hydrothermal	1.0.14 / 2011		77.51	no change in	[15]
		1.0 M KOH	18140		the CV curves	
PdZn		and 1.0M			after 2000	
		ethanol			cycles	
PdSn/rgo	Seed-mediated	1.0 M KOH			1187 mA mg <sup>-1</sup>	
		and 1.0M	2701	23.6	retained after	[16]
		ethanol			500 cycles	
		1.0 M KOH			87.6% after	
PdNN	Hydrothermal	and 1.0M	2040	100	the 1000th	[17]
		ethanol			cycle	
	Co-reduction at room temperature	1.0 M KOH			1 ( 0 0 1 1	<b>T</b> 1 :
PdSn/C-1		and 1.0 M	8452	104.3	after 1500 s	This
		ethanol				work

## Table S3. The fitted parameters in equivalent circuit model of catalysts

Nanoparticles	$R_{ct}\left(\Omega ight)$	$\mathrm{R}_{\mathrm{s}}\left(\Omega ight)$
PdSn NPs-0.1	385	12.3
PdSn NPs-0.5	139	14.1
PdSn NPs-1	83	11.5
PdSn NPs-2	165	17.0
Pd/C	882	14.4

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