Energy Materials

Supplementary Materials

Multifunctional zinc silicate coating layer for high-performance aqueous zinc-ion batteries

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Reactions procedures to form Zinc silicate polymer protection film:





Figure 1. SEM images of (A, B) Zn-LSO-0.5; (D, E) Zn-LSO-1.5 with different magnification and cross-section of (C) Zn and Zn-LSO-0.5; (F) Zn and Zn-LSO-1.5.



Figure 2. Elements mapping of Zn-LSO-1.



Figure 3. Tafel curves of Zn, Zn-LSO-0.5 and Zn-LSO-1.5 anode under the three-electrode system.



Figure 4. Nyquist plots of Zn-LSO-0.5 and Zn-LSO-1.5 anode.



Figure 5. Galvanostatic cycling stability at a current density of 3 mA cm⁻² with an areal capacity of 0.1 mAh cm⁻² for symmetrical (A) Zn and Zn-LSO-0.5; (B) Zn and Zn-LSO-1.5 cells, and initial voltage profiles of symmetrical (C) Zn and Zn-LSO-0.5; (D) Zn and Zn-LSO-1.5 cells.



Figure 6. (A) X-ray diffraction of MnO2 Zn-LSO-1.5; (B) SEM images of MnO₂.



Figure 7. (A) galvanostatic charge and discharge potential profiles of (B) Zn/MnO₂ and (d) Zn-LSO-1/MnO₂ cells at different current densities.



Figure 8. Rate performance of Zn/MnO₂ and Zn-LSO-1/MnO₂ cells from 0.2 to 2 A g⁻¹.

Electrode	Electrolyt e	Capacity (mAh cm ⁻²)	Current density (mA cm ⁻ ²)	Voltage hysteresis (mV)	Cycling life (h)	Referenc e
Graphite- coated Zn anode	2 M ZnSO4	0.1	0.1	28	200	[1]
Nanoporou s CaCO ₃	3 M ZnSO ₄ + 0.1 M MnSO ₄	0.05/0.1/0.1/ 0.1	0.25/1/2/3	80/140/200 /160	836/80/80/80	[2]
Kaolin- coated Zn anode	2 M ZnSO ₄ + 0.1 M MnSO ₄	1.1	4.4	70	800	[3]
PVB@Zn	ZnSO ₄	0.5	0.5	108.6	2200	[4]
Zn/CNT	2 M ZnSO4	2	2	27	200	[5]
MXene@Z n Paper	2 M ZnSO ₄	1	1	75	300	[6]
100TiO ₂ @ Zn	3 M Zn(SO ₃ CF 3)2	1	1	57.2	150	[7]
Zn/rGO	2 M ZnSO ₄ + 0.1 M MnSO ₄	1/2	1	60/100	200	[8]

Table 1. Comparison of lifespan of Zn-LSO-1 with reported Zn anodes
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Zn-LSO-1	2 M								
	$ZnSO_4 +$	0.1	0.5/1/3/5	66/80/118/	835/455/344/	This work			
	0.1 M			141	260	THIS WOLK			
	MnSO ₄								

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